

Gas Laws Practice

Name: _____

Date: _____

Hour: _____

IMPORTANT: whenever you use temperature, it must be in degree Kelvin (K), so remember the equation: $K = ^\circ C + 273$

1. a) convert $39\text{ }^\circ\text{C}$ to K. b) convert 127 K to $^\circ\text{C}$.
2. A gas has an initial volume of 2.75 L at a temperature of 285 K . If the temperature changes to 380 K , what is the new volume of the gas if the pressure is unchanged?
3. Gas can often be cooled by compressing it while keeping the pressure constant. If I have 45.0 L of gas at room temperature (22°C) and I compress it so that the final volume is 0.50 L , what is the final temperature of the gas if the pressure is constant?
4. The volume of a gas is 2.5 L when the pressure is at standard pressure (101.325 kPa). What is the volume of the gas if the pressure decreases to 85 kPa and the temperature remains unchanged?
5. A 5.0 L container of gas experiences a temperature change so that the final temperature is 4 times the initial temperature. What is the size of the container after the temperature change? (Assume constant pressure.)
6. At 45°C the volume of a certain gas is 27.5 L and the pressure is 210 kPa . What is the volume of the gas at standard temperature (273 K) and 310 kPa of pressure?
7. The pressure of a sample of gas was 97.8 kPa and the volume of the gas was 3.75 L . If the gas occupied a container with a volume of 8.00 L , what would the pressure in the container be?
8. Isothermal expansion refers to allowing a gas to expand while keeping the temperature constant. This is one means to simulate a vacuum. If a gas originally at 97 kPa is allowed to expand from 0.25 L to 182 L , what is the pressure of the gas?
9. A gas is initially at a pressure of 225 kPa and a temperature of 245 K in a container that is 4.5 L . If the gas is compressed to a volume of 2.1 L and the temperature changes to 275 K , what is the new pressure?

Skill Practice 35 Gas Law Practice Answers

Keith Johnson



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Chemfile Skills Practice Experiments Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2006 *West's Federal Practice Digest 3d*, 1984 *West's Federal Practice Digest 2d*, 1976 *West's Federal Practice Digest 4th*, 2002 Locate federal cases decided in the U S Supreme Court Court of Appeals district courts Claims Court bankruptcy courts Court of Military Appeals the Courts of Military Review and other federal courts This Key Number Digest contains all headnotes classified according to West's Key Number System for federal court decisions reported from 1984 to the present The topics are listed in alphabetical order The Key Numbers within those topics are listed in numerical order Each topic begins with scope notes about subjects included and subjects excluded and covered by other topics Also there is an outline of the topic which includes a list of all Key Numbers in that topic Headnotes are collected by jurisdiction or court and filed according to the West Key Number System **West's Federal Practice Digest**, 2002 **New Physics for You** Keith Johnson, 2001 for

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