

- The particles in a gas are more spread out than the particles in a liquid and have less attractive forces acting between them; so, if the same amount of energy is transferred to a gas as to a liquid, the kinetic energy of the particles in a gas will increase more than the kinetic energy of particles in a liquid.
- Convection—movement of the hot water as it expands and rises, then cools and contracts, mixes the hot and cold water.
- Students should defend their answers. A heavy cast iron skillet transfers heat evenly, resists temperature changes, and is slow to heat. A thin stainless steel skillet transfers heat quickly, has a high surface temperature directly over the flame, and cools quickly.

SECTION: USING HEAT

- A damp towel has water that absorbs heat energy and evaporates as the air blows through it. Because the air loses heat energy to the water, the air is cooler.
- No, the refrigerator releases heat from inside the box out the back of the unit. The refrigerator is in the room, so the heat is still released in the kitchen.
 - Yes, because shivering causes muscle movements that generate heat.
 - Yes, because the large ears of a jackrabbit act as radiators to radiate heat from the rabbit's blood.
- The two objects that will be at different temperatures, such as an inside wall and an outside wall, should be separated by an air-filled space so that they are not touching, in order to eliminate heat loss by conduction.
- A diesel engine does not use spark plugs. Instead, it compresses the fuel-air mixture so much that a spark is ignited.
- $$T_F = 1.8t + 32.0 = [1.8(-235)] + 32.0 = -391^\circ\text{F}$$

$$T = t + 273 = -235 + 273 = 38\text{ K}$$
- $$T_F = 1.8t + 32.0 = (1.8 \times 15) + 32.0 = 59^\circ\text{F}$$

$$T = t + 273 = 15 + 273 = 288\text{ K}$$
- $$T_F = 1.8t + 32.0 = (1.8 \times 190) + 32.0$$

$$T = t + 273 = 190 + 273 = 463\text{ K}$$

$$= 370^\circ\text{F}$$
- $$t = \frac{(T_F - 32.0)}{1.8} = \frac{(125 - 32.0)}{1.8} = 52^\circ\text{C}$$

$$T = t + 273 = 52 + 273 = 325\text{ K}$$
- $$t = \frac{(T_F - 32.0)}{1.8} = \frac{(17 - 32.0)}{1.8} = -8.3^\circ\text{C}$$

$$T = t + 273 = 265\text{ K}$$

$$t = \frac{(T_F - 32.0)}{1.8} = \frac{(83 - 32.0)}{1.8} = 28^\circ\text{C}$$

$$T = t + 273 = 28 + 273 = 301\text{ K}$$
- $$t = \frac{(T_F - 32.0)}{1.8} = \frac{(-83.0 - 32.0)}{1.8}$$

$$= -64.4^\circ\text{C}$$

$$T = t + 273 = -64.4 + 273 = 209\text{ K}$$
- $$t = \frac{(T_F - 32.0)}{1.8} = \frac{(276 - 32.0)}{1.8}$$

$$= 136^\circ\text{C}$$

$$T = t + 273 = 136 + 273 = 409\text{ K}$$
- $$t = T - 273 = 5100 - 273 = 4800^\circ\text{C}$$

$$T_F = 1.8t + 32.0 = 1.8(4800) + 32.0 = 8700^\circ\text{F}$$
- $$t = T - 273 = 1088 - 273 = 815^\circ\text{C}$$

$$T_F = 1.8t + 32.0 = 1.8(815) + 32.0 = 1.50 \times 10^3^\circ\text{F}$$
- $$t = T - 273 = 3500 - 273 = 3200^\circ\text{C}$$

$$T_F = 1.8t + 32.0 = 1.8(3200) + 32.0 = 5800^\circ\text{F}$$
- $$t = T - 273 = 240 - 273 = -30^\circ\text{C}$$

$$T_F = 1.8t + 32.0 = 1.8(-30) + 32.0 = -20^\circ\text{F}$$
- $$T_F = 1.8t + 32.0 = 1.8(29) + 32.0 = 84^\circ\text{F}$$

$$T_F = 1.8t + 32.0 = 1.8(-11) + 32.0 = 12^\circ\text{F}$$
- $$t = T - 273 = 1.0 \times 10^5 - 273 = 1.0 \times 10^5\text{C}$$
- $$t = T - 273 = 4.25 - 273 = -269^\circ\text{C}$$

$$T_F = 1.8t + 32.0 = 1.8(-269) + 32.0 = -482^\circ\text{F}$$
- $$\Delta t = \frac{(\Delta T_F)}{1.8} = \frac{(49^\circ\text{F})}{1.8} = 27^\circ\text{C}$$
 - $$\text{initial } t = \text{final } t - \Delta t$$

$$= 45^\circ\text{F} - 49^\circ\text{F} = -4^\circ\text{F}$$

Math Skills

TEMPERATURE CONVERSION

- $$T_F = 1.8t + 32.0 = (1.8 \times 453) + 32.0 = 847^\circ\text{F}$$

$$T = t + 273 = 453 + 273 = 726\text{ K}$$

Skills Math Skills Specific Heat Answers

JL Elias



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Math Skills for the Sciences John G. Pearson, Dennis M. Stone, Richard F. Swindell, 1976 *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science*, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Science Spectrum** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03

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