

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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Table of Contents Skills Concept Review Answer Key Holt Chemistry

1. Understanding the eBook Skills Concept Review Answer Key Holt Chemistry
 - The Rise of Digital Reading Skills Concept Review Answer Key Holt Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Skills Concept Review Answer Key Holt Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Skills Concept Review Answer Key Holt Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Skills Concept Review Answer Key Holt Chemistry
 - Personalized Recommendations
 - Skills Concept Review Answer Key Holt Chemistry User Reviews and Ratings
 - Skills Concept Review Answer Key Holt Chemistry and Bestseller Lists
5. Accessing Skills Concept Review Answer Key Holt Chemistry Free and Paid eBooks
 - Skills Concept Review Answer Key Holt Chemistry Public Domain eBooks
 - Skills Concept Review Answer Key Holt Chemistry eBook Subscription Services
 - Skills Concept Review Answer Key Holt Chemistry Budget-Friendly Options
6. Navigating Skills Concept Review Answer Key Holt Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Skills Concept Review Answer Key Holt Chemistry Compatibility with Devices
 - Skills Concept Review Answer Key Holt Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Skills Concept Review Answer Key Holt Chemistry
 - Highlighting and Note-Taking Skills Concept Review Answer Key Holt Chemistry
 - Interactive Elements Skills Concept Review Answer Key Holt Chemistry
8. Staying Engaged with Skills Concept Review Answer Key Holt Chemistry

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Skills Concept Review Answer Key Holt Chemistry
- 9. Balancing eBooks and Physical Books Skills Concept Review Answer Key Holt Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Skills Concept Review Answer Key Holt Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Skills Concept Review Answer Key Holt Chemistry
 - Setting Reading Goals Skills Concept Review Answer Key Holt Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Skills Concept Review Answer Key Holt Chemistry
 - Fact-Checking eBook Content of Skills Concept Review Answer Key Holt Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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