

Chapter 5 The Periodic Table

Section 5.2 The Modern Periodic Table**(pages 130–138)**

This section explains the organization of the modern periodic table and discusses the general properties of metals, nonmetals, and metalloids.

Reading Strategy (page 130)

Previewing Before you read, complete the table by writing two questions about the periodic table on pages 132–133. As you read, write answers to your questions. For more information on this reading strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Questions About the Periodic Table	
Question	Answer

The Periodic Law (pages 131–133)

1. Is the following sentence true or false? In the modern periodic table, elements are arranged by increasing number of protons.

2. Explain why the number of elements per period varies. _____

3. Properties of elements repeat in a predictable way when atomic numbers are used to arrange elements into groups. This pattern of repeating properties is called the _____.

Atomic Mass (page 134)

4. Label the four types of information supplied for chlorine in the diagram.

a. _____	17
b. _____	Cl
c. _____	Chlorine
d. _____	35.453

- a. _____ b. _____
c. _____ d. _____

Skills Practice 1 Prentice Hall Answers

Iztok Devetak, Saša Aleksij Glažar



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Writing and Grammar: Communication in Action Pearson/Prentice Hall, 2003-05 *Standardized test preparation workbook*, 2002 **Prentice Hall Math Course 3 Adapted Student Workbook 2007c** Prentice-Hall Staff, 2006-07-15
Prentice Hall Mathematics Course 3 A solid foundation preparing students for Algebra 1 Chapters 1-3 provide a solid foundation of integers rational numbers and real numbers setting the stage for equations inequalities and functions Real World applications to the more abstract algebraic concepts are found throughout the text An average of over five Activity Labs per chapter ensures students receive the visual and special instruction necessary to conceptualize these abstract concepts better preparing them for advanced math courses EPIE Educational Product Report EPIE Institute, 1974

Skills Intervention Unit Pre-Algebra Basics Wkbk 2001c Prentice Hall, 2000-05 We are pleased to offer you and your students these economical Value Pack combinations for the Mathematics classroom We've assembled our most popular student resources to bring you a variety of ways to integrate programs seamlessly at a substantial savings Pearson Prentice Hall Value Packs make the most of dollars and sense *Literature Standardized Test Preparation : Silver Level* □□ □, 2005-01-10 Handbook for Teaching Introductory Psychology Michelle Rae Hebl, Charles L. Brewer, Ludy T. Benjamin, Jr., 2001-08-01 Like its predecessors Volume III of the Handbook for Teaching Introductory Psychology provides introductory psychology instructors with teaching ideas and activities that can immediately be put into practice in the classroom It contains an organized collection of articles from Teaching of Psychology TOP the official journal of the Society for the Teaching of Psychology Division 2 of the American Psychological Association Volume III contains 89 articles from TOP that have not been included in other volumes Another distinction between this volume and its predecessors is its emphasis on testing and assessment The book is divided into two sections Section One Issues and Approaches in Teaching Introductory Psychology contains 52 articles on critical issues such as how to approach the course understanding students interests perceptions and motives students existing knowledge of psychology including their misconceptions a comparison of introductory textbooks and tips on how to evaluate them test questions and student factors affecting exam performance an overview of different forms of feedback giving extra credit and how to deal with academic dishonesty Section Two consists of 37 articles that present demonstrations class and laboratory projects and other techniques to enhance teaching and learning in both the introductory as well as advanced courses in the discipline This section is organized so as to parallel the order of topics found in most introductory psychology textbooks Intended for academicians who teach the introductory psychology course and or oversee grad assistants who teach the course all royalties of the book go directly to the Society for the Teaching of Psychology to promote its activities to further improve the teaching of psychology *Learning with Understanding in the Chemistry Classroom* Iztok Devetak, Saša Aleksij Glažar, 2014-01-14 This volume offers a critical examination of a variety of conceptual approaches to teaching and learning chemistry in the school classroom Presenting up

to date research and theory and featuring contributions by respected academics on several continents it explores ways of making knowledge meaningful and relevant to students as well as strategies for effectively communicating the core concepts essential for developing a robust understanding of the subject Structured in three sections the contents deal first with teaching and learning chemistry discussing general issues and pedagogical strategies using macro sub micro and symbolic representations of chemical concepts Researchers also describe new and productive teaching strategies The second section examines specific approaches that foster learning with understanding focusing on techniques such as cooperative learning presentations laboratory activities multimedia simulations and role playing in forensic chemistry classes The final part of the book details learner centered active chemistry learning methods active computer aided learning and trainee chemistry teachers use of student centered learning during their pre service education Comprehensive and highly relevant this new publication makes a significant contribution to the continuing task of making chemistry classes engaging and effective

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