

States of Matter

Review and Reinforce

1. The bonds of cohesiveness would be relaxed and so a liquid, and the vapors so as gas.
2. The solid would lose the water drops because the particles in a solid are bound tightly together. The drops of the liquid will be able to spread and later evaporate because the particles in a liquid have greater freedom. The drops of the vapors would continue to change and spread without because the particles in a gas move apart in many different directions.
3. Since the bonds of cohesiveness which the volume would stop the water. The volume of the liquid will be smaller than the water. The volume of the vapors would expand to fill the area of the larger container because the particles would spread apart.
4. Matter with a definite volume and a definite shape.
5. Matter with a definite volume but no shape will be very.
6. Matter without a definite shape or volume.
7. The molecules of a liquid in flowing.
8. It refers to where the particles are not arranged in a regular pattern.
9. It refers to where the particles are arranged in a regular repeating pattern.
10. A substance that flows.
11. The bonds of cohesiveness will prevent the molecules of a liquid from being farther apart than they are together.

Changes of State

Review and Reinforce

1. Gases, liquids
2. Gases, liquids
3. Gases, liquids
4. Gases, liquids
5. Gases, liquids
6. boiling point
7. condensation
8. evaporation
9. boiling
10. evaporation
11. melting
12. freezing
13. melting point
14. sublimation

Skills Reinforcement Activity 20 Answers

Yi-Tong Ma



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