

GCSE CHEMISTRY PAPER 2 AQA REVISION NOTES

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GCSE CHEMISTRY: ATOMIC STRUCTURE

Atomic Structure

Atoms are made of three particles: protons, neutrons and electrons.

Protons and neutrons are found in the nucleus. Electrons are found in shells.

The mass of an atom is the sum of the number of protons and neutrons.

The relative atomic mass (A_r) is the mass of an atom relative to $\frac{1}{12}$ th the mass of a carbon-12 atom.

Periodic Table

The periodic table is a table of elements arranged in order of increasing atomic number.

Elements are grouped into metals, non-metals and metalloids.

The periodic table shows the relationship between the elements.

Bonding

Atoms bond together to form molecules or giant structures.

There are three types of bonding: ionic, covalent and metallic.

Chemical Reactions

Chemical reactions involve the breaking and making of bonds.

Chemical reactions can be exothermic or endothermic.

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