

Name:		
	Date:	

Hour: ___

I. Single Replacement Reactions

(Remember, some reactions will NOT proceed. If there is no reaction, write "NR" to the right of the arrow.)

- Write the reaction for each of the following reactions:
 - a) Sodium sulfate reacts with magnesium to produce magnesium sulfate and sodium
 - b) Copper(I) carbonate reacts with aluminum to form copper metal and aluminum carbonate.
- 2. Complete the following reactions:
 - a) Ca(NO₃)₂ + AI →
 - b) Zn + K₃PO₄ →

II. Double Replacement Reactions

- 3. Write the reaction for each of the following reactions:
 - a) Iron(II) phosphate reacts with sodium chloride to produce iron(II) chloride and sodium phosphate.
 - b) Calcium carbonate reacts with lithium hydroxide to produce calcium hydroxide and lithium carbonate.
- 4. Complete the following reactions:
 - a) Mg(NO₃)₂ + Al(C₂H₃O₂)₅ →
 - b) Ba(OH)₂ + Na₃PO₄ →

III. Combustion Reactions

You can try the following formula to balance: $(CxHy + (x + y/4)O_2 \rightarrow xCO_2 + yH_2O)$

- Write the combustion reactions for the following compounds.
 - a) C₂H₄
 - b) C₈H₁₆

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