

Skills Worksheet

Math Skills

Determining pH

After you study each sample problem and solution, work out the practice problems on a separate sheet of paper. Write your answers in the spaces provided.

PROBLEM

What is the pH of a 0.001 M solution of the strong acid HI dissolved in water?

SOLUTION

Step 1: List the given and unknown values.

Given: concentration of HI in solution = 0.001 M

Unknown: pH

Step 2: Determine the molar concentration of hydroxide ions.

concentration of HI in solution = 0.001 M

HI is a strong acid, so the concentration of hydroxide ions in the solution is equal to the concentration of HI

concentration of H_3O^+ ions = concentration HI = 0.001 M = 1×10^{-3} M

Step 3: Convert the H_3O^+ concentration to pH

pH = the negative of the power of ten used to describe the concentration of H_3O^+ ions

concentration of H_3O^+ ions = 1×10^{-3} M

pH = $-(-3) = 3$

PRACTICE

1. A strong acid, HBr, has been dissolved into a beaker of water. If the solution is known to be 0.000 001 M, what is the pH of the solution?

Skills Math Skills Determining Ph

Jonathan Duhamel



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Table of Contents Skills Math Skills Determining Ph

1. Understanding the eBook Skills Math Skills Determining Ph
 - The Rise of Digital Reading Skills Math Skills Determining Ph
 - Advantages of eBooks Over Traditional Books
2. Identifying Skills Math Skills Determining Ph
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Skills Math Skills Determining Ph
 - User-Friendly Interface
4. Exploring eBook Recommendations from Skills Math Skills Determining Ph
 - Personalized Recommendations
 - Skills Math Skills Determining Ph User Reviews and Ratings
 - Skills Math Skills Determining Ph and Bestseller Lists

5. Accessing Skills Math Skills Determining Ph Free and Paid eBooks
 - Skills Math Skills Determining Ph Public Domain eBooks
 - Skills Math Skills Determining Ph eBook Subscription Services
 - Skills Math Skills Determining Ph Budget-Friendly Options
6. Navigating Skills Math Skills Determining Ph eBook Formats
 - ePub, PDF, MOBI, and More
 - Skills Math Skills Determining Ph Compatibility with Devices
 - Skills Math Skills Determining Ph Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Skills Math Skills Determining Ph
 - Highlighting and Note-Taking Skills Math Skills Determining Ph
 - Interactive Elements Skills Math Skills Determining Ph
8. Staying Engaged with Skills Math Skills Determining Ph
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Skills Math Skills Determining Ph
9. Balancing eBooks and Physical Books Skills Math Skills Determining Ph
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Skills Math Skills Determining Ph
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Skills Math Skills Determining Ph
 - Setting Reading Goals Skills Math Skills Determining Ph
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Skills Math Skills Determining Ph
 - Fact-Checking eBook Content of Skills Math Skills Determining Ph
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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