# Name Key

# Concentration Practice Problems

#### % Solution

 What is the concentration (% solution) of a solution that contains 0.78 grams of MgCl<sub>2</sub> dissolved in 10, grams of water?

$$\frac{970}{\text{mass of soluti}} = \frac{0.789}{10.789} = \frac{7.24976}{10.789}$$

How many grams of solid are needed to prepare a 10.0% NaCl solution in 200, grams of water?

What is the % solution of a mixture that contains 0.0500 moles of NaI dissolved in 25.0 g of

#### Molarity

 What is the molarity of a solution that contains 85.0 q of iodine (I<sub>2</sub>) dissolved in 250. mL of ethanol solution?

5. How many grams of Ca(NO<sub>3</sub>)<sub>2</sub> are need to produce 1500. mL of a 0.75 M solution?

$$M = \frac{mol}{L}$$

$$0.75 = \frac{x}{1.500 L}$$

$$x = 1.125 \text{ mol}$$

$$1.125 \text{ mol}$$

$$1.125 \text{ mol}$$

$$1.125 \text{ mol}$$

$$1.125 \text{ mol}$$

# **Skills Practice 39 Concentration Practice Answers**

**Kelly Hackett** 

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