

Student Lab Handout



Deep Frying Direction



Introduction

The purpose of this laboratory is to determine the effect of temperature on the rate of a chemical reaction. The reaction is the decomposition of hydrogen peroxide into water and oxygen. The reaction is catalyzed by the enzyme catalase. The reaction is exothermic and the rate of reaction is affected by temperature.

Review Labels

What are the labels for the following reactions?

- | | |
|---|------------|
| 1. Decomposition of hydrogen peroxide | Exothermic |
| 2. Combustion of the candle | Exothermic |
| 3. Synthesis of water from hydrogen and oxygen | Exothermic |
| 4. The acid-base reaction of acetic acid and sodium hydroxide | Exothermic |
| 5. Synthesis of water from hydrogen and oxygen | Exothermic |
| 6. Decomposition of hydrogen peroxide | Exothermic |
| 7. Synthesis of water from hydrogen and oxygen | Exothermic |
| 8. Synthesis of water from hydrogen and oxygen | Exothermic |
| 9. Synthesis of water from hydrogen and oxygen | Exothermic |
| 10. Synthesis of water from hydrogen and oxygen | Exothermic |

Materials

Hydrogen peroxide
Catalase

Hydrogen peroxide
Catalase



Sheep Kidney Lab Answer Key

Jeffrey C. Pommerville

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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