

PROPERTIES OF WATER

Reading Comprehension

Grades 8-12

NAME: _____ **PER:** _____ **DATE:** _____

PROPERTIES OF WATER

Heat Capacity and High Heat of Vaporization

(1) Heat capacity refers to the ability of a substance to absorb heat. Water has a high specific heat capacity, meaning it can absorb a lot of heat without its temperature rising very quickly. This means that water can absorb a lot of heat from the sun or a fire without getting too hot. This property is what allows water to regulate Earth's climate. For example, the ocean's high heat capacity helps to keep the temperature of the air above it from rising too quickly. Additionally, water's high heat of vaporization means that it takes a lot of energy to turn liquid water into a gas. This property is what allows water to cool the body when it evaporates. For example, when you sweat, the water on your skin evaporates, taking heat with it and cooling your body.

(2) Water has a high heat of vaporization, meaning it takes a lot of energy to turn liquid water into a gas. This property is what allows water to cool the body when it evaporates. For example, when you sweat, the water on your skin evaporates, taking heat with it and cooling your body. Additionally, water's high heat of vaporization means that it takes a lot of energy to turn liquid water into a gas. This property is what allows water to cool the body when it evaporates. For example, when you sweat, the water on your skin evaporates, taking heat with it and cooling your body.



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PROPERTIES OF WATER

Adhesion and Cohesion

(1) Hydrogen bonds between water molecules give water the properties of adhesion and cohesion. Adhesion is the attraction between water molecules and other substances. Cohesion is the attraction between water molecules and other water molecules. This property is what allows water to move up a plant stem. For example, water molecules are attracted to the walls of the xylem vessels in a plant stem, allowing them to move up. Additionally, water molecules are attracted to each other, allowing them to move up. This property is what allows water to move up a plant stem.

(2) Cohesion occurs because of the hydrogen bonds between water molecules. This property is what allows water to move up a plant stem. For example, water molecules are attracted to each other, allowing them to move up. Additionally, water molecules are attracted to the walls of the xylem vessels in a plant stem, allowing them to move up. This property is what allows water to move up a plant stem.

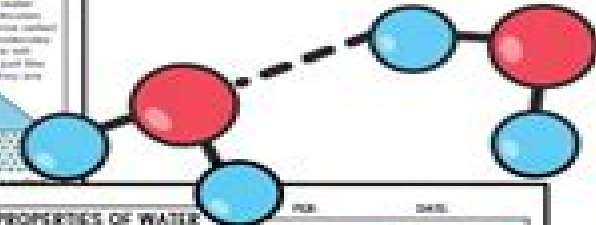



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PROPERTIES OF WATER

Surface Tension

(1) Because of water's cohesive nature, each molecule of water has two hydrogen bonds with other water molecules. This strong attraction between water molecules produces a force called surface tension. Surface tension is the force that holds the molecules of water together at the surface. This property is what allows water to move up a plant stem. For example, water molecules are attracted to each other, allowing them to move up. Additionally, water molecules are attracted to the walls of the xylem vessels in a plant stem, allowing them to move up. This property is what allows water to move up a plant stem.

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PROPERTIES OF WATER

Hydrogen Bonds

(1) Water molecules are made of two hydrogen atoms and one oxygen atom. The oxygen atom is more electronegative than the hydrogen atoms, meaning it has a stronger attraction for the shared electrons in the bonds. This creates a partial negative charge on the oxygen atom and a partial positive charge on the hydrogen atoms. This property is what allows water to move up a plant stem. For example, water molecules are attracted to each other, allowing them to move up. Additionally, water molecules are attracted to the walls of the xylem vessels in a plant stem, allowing them to move up. This property is what allows water to move up a plant stem.

(2) The polarity of water molecules allows them to move up a plant stem. For example, water molecules are attracted to each other, allowing them to move up. Additionally, water molecules are attracted to the walls of the xylem vessels in a plant stem, allowing them to move up. This property is what allows water to move up a plant stem.




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The Properties Of Water Guided Reading Study Answer Key

Rachel Sandford



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