Describe the differences between alkali metals, alkaline earth metals, and transition metals. Provide examples of each and discuss their uses.

Answer Key Multiple Choice

1 0 2.0 3.0 44 . . . 5. a 6. c 7. b B. C. 916 1.0 11. b 12. b. 1.3 3 14 C 15. 1.6 O. 17. 18 b 19. C. 20. b 21. 100 22 23 8 24. by 25. a.

Essay Questions

- Metallic bonding involves a "sea" of delocalized electrons surrounding positively charged metal ions.²
 - Conductivity: Free electrons can move and carry electrical current.³

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Y Pai

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