#### ACS Final Exam Practice 1.

The exam is 70 questions in 110 minutes. You may use the equation sheet and periodic table but no other materials

#### Section 1 Foundations

PQ-1. The speed of light is 3.0 × 10 m·s 1. What is this speed in nm·ms 1?

(A) 3.0×10<sup>-1</sup> nm-ms<sup>-1</sup>

(B) 3.0×10<sup>2</sup> nm·ms <sup>1</sup>

(C) 3.0×10<sup>-4</sup> nm-ms<sup>-1</sup> (D) 3.0×10<sup>20</sup> nm-ms<sup>-1</sup>

Mass of beaker and water

PO-4. What is the correctly reported mass of water based on the data in the salde?

(A) 1.3 g

1.30 g

Mass of bealer only

(C) 1,298 g

(D) 1.2980 g

29.62 亩

28.3220 m

PQ-5. What is the name of Ti<sub>2</sub>(PO<sub>4</sub>)<sub>4</sub>?

titanium phosphate

citanium(IV) phosphate

(B) titanium(III) phosphate

(D) titanium tetruphosphate

PQ-6. The formula of strontium bexafluorosilicate is SrSil's. What is the formula of aluminum hexafluorosilicate?

(A) AlSiFa

Alp(SiFa)s

(C) AbSiFe

(D) Ab(SiFa):

PO-8. What is the correct name for N-O-7.

(A) nitrogen(II) oxide

(C) dinitrogen oxide

mitrogen(V) oxide

dinitrogen pentoxide

PQ-12. If a palladium nanoparticle has a density of 12.0 g-cm<sup>-3</sup>, what is the mass of a nanoparticle with a volume of 1.84×10-21 cm3?

2.21×10-20 g

(B) 1.63×10<sup>-30</sup> g

(C) 6.52×10<sup>-21</sup> at

(D) 1.53×10<sup>-22</sup> g

# **Standardized Organic Chemistry Exam**

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web directions for dissection on the back side of this sheet fill in a box with the name how many and function of each part color code the box to match the flower part 1 find and

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