

4. A solution contains 2.0 g of  $\text{PbCl}_2$  in 100 mL of water. The solution is heated to  $100^\circ\text{C}$ . The percentage of  $\text{PbCl}_2$  that remains dissolved at  $100^\circ\text{C}$  is:  
a. 10.0%  
b. 10.7%  
c. 10.8%  
d. 10.9%
5. A solution contains 10.0 g of  $\text{PbCl}_2$  in 100 mL of water. The solution is heated to  $100^\circ\text{C}$ . The percentage of  $\text{PbCl}_2$  that remains dissolved at  $100^\circ\text{C}$  is:  
a. 10.0%  
b. 10.7%  
c. 10.8%  
d. 10.9%
6. The percentage of the compound that remains dissolved at  $100^\circ\text{C}$  is:  
a. 10.0%  
b. 10.7%  
c. 10.8%  
d. 10.9%
7. A solution contains 10.0 g of  $\text{PbCl}_2$  in 100 mL of water. The solution is heated to  $100^\circ\text{C}$ . The percentage of  $\text{PbCl}_2$  that remains dissolved at  $100^\circ\text{C}$  is:  
a. 10.0%  
b. 10.7%  
c. 10.8%  
d. 10.9%
8. The percentage of the compound that remains dissolved at  $100^\circ\text{C}$  is:  
a. 10.0%  
b. 10.7%  
c. 10.8%  
d. 10.9%
9. A solution contains 10.0 g of  $\text{PbCl}_2$  in 100 mL of water. The solution is heated to  $100^\circ\text{C}$ . The percentage of  $\text{PbCl}_2$  that remains dissolved at  $100^\circ\text{C}$  is:  
a. 10.0%  
b. 10.7%  
c. 10.8%  
d. 10.9%
10. A solution contains 10.0 g of  $\text{PbCl}_2$  in 100 mL of water. The solution is heated to  $100^\circ\text{C}$ . The percentage of  $\text{PbCl}_2$  that remains dissolved at  $100^\circ\text{C}$  is:  
a. 10.0%  
b. 10.7%  
c. 10.8%  
d. 10.9%

# Standardized Test Practice Chapter 13 Chemistry Answers

**Amanda Stewart**



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**Glencoe Science** McGraw-Hill Staff, 2001-06

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