

Solution Manual

Section 1.1

- | | |
|-----------------------------|-----------------------------|
| 1. first-order, linear | 2. first-order, nonlinear |
| 3. first-order, nonlinear | 4. third-order, linear |
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Section 1.2

1. Because the differential equation can be rewritten $e^{-y} dy = x dx$, integration immediately gives $-e^{-y} = \frac{1}{2}x^2 + C$, or $y = -\ln(C - x^2/2)$.
2. Separating variables, we have that $dx/(1+x^2) = dy/(1+y^2)$. Integrating this equation, we find that $\tan^{-1}(x) - \tan^{-1}(y) = \tan(C)$, or $(x-y)/(1+xy) = C$.
3. Because the differential equation can be rewritten $\ln(x)dx/x = y dy$, integration immediately gives $\frac{1}{2}\ln^2(x) + C = \frac{1}{2}y^2$, or $y^2(x) - \ln^2(x) = 2C$.
4. Because the differential equation can be rewritten $y^2 dy = (x+x^3) dx$, integration immediately gives $y^3(x)/3 = x^2/2 + x^4/4 + C$.
5. Because the differential equation can be rewritten $y dy/(2+y^2) = x dx/(1+x^2)$, integration immediately gives $\frac{1}{2}\ln(2+y^2) = \frac{1}{2}\ln(1+x^2) + \frac{1}{2}\ln(C)$, or $2+y^2(x) = C(1+x^2)$.
6. Because the differential equation can be rewritten $dy/y^{1/3} = x^{1/3} dx$, integration immediately gives $\frac{3}{2}y^{2/3} = \frac{3}{2}x^{4/3} + \frac{3}{2}C$, or $y(x) = (\frac{1}{2}x^{4/3} + C)^{3/2}$.

Solutions Manual Vanloon Duffy

Gary W. VanLoon, Stephen J. Duffy



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Gary W. VanLoon, Stephen J. Duffy, 2005 Guiding us through the chemical composition of the three key environmental systems the atmosphere hydrosphere and terrestrial environment the authors explain the chemical processes which occur within and between each system Focusing on general principles we are introduced to the essential chemical concepts which underpin an understanding of the air water and soil and how they behave careful explanations ensure that clarity is not sacrificed at the expense of thorough coverage of the underlying chemistry We then see how human activity continues to affect the chemical behavior of these environmental systems and what the consequences of these natural processes being disturbed can be *Environmental Chemistry A Global Perspective* takes chemistry out of the laboratory and shows us its importance in the world around us With illuminating examples from around the globe its rich pedagogy and broad carefully structured coverage this book is the perfect resource for any environmental chemistry student wishing to develop a thorough understanding of their subject Supplementary Resources r Companion website featuring downloadable illustrations Solutions manual **Forthcoming Books** Rose Arny, 1999 **Bookseller and Stationer** ,1922 Indian Books in Print ,2003

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