

## Solutions, Suspensions, and Colloids Lab

**Objective:** To observe various mixtures and identify them as a solution, suspension, or colloid

**Material:** sucrose, soluble starch, clay, food coloring, salt, cooking oil, gelatin, vinegar, flashlight

**Theory:** Three different types of mixtures can be identified by using a beam of light to illustrate the Tyndall Effect. If the mixture is transparent and the Tyndall Effect is not seen, the mixture is a solution. If the mixture shows a Tyndall Effect and does not settle, the mixture is a colloid. If the mixture shows the Tyndall Effect and settles, it is a suspension. *(When a beam of light passes through a colloid, the beam is visible and appears as a ray passing through the material. Inside the container of a colloid, a light beam is visible as a set path. When a beam of light passes through a true solution, the beam is not visible when passing through the material. Inside the container of a solution, the beam is not visible as a set path.)*



**Procedure:**

1. Describe each mixture by placing your observations in the data table under "appearance".
2. Shine a flashlight through each mixture and observe what happens. The Tyndall Effect will either be present or not present. (Refer to the theory section above for further explanation.)
3. Based on procedure steps 1 and 2, classify the mixtures as solutions, suspensions, or colloids. (Refer to the theory section above for further explanation.)
4. Complete the reasoning column in the data table by explaining why you chose the classification that you did for each mixture.

**Observation Table:**

Mixture	Appearance	Tyndall Effect?	Classification	Reasoning
Sucrose + H <sub>2</sub> O				
Starch + H <sub>2</sub> O				
Clay + H <sub>2</sub> O				
Food Coloring + H <sub>2</sub> O				
Salt Water				
Cooking Oil				

# Solutions Suspensions And Colloids Lab Answer Key

**SB Merriam**



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**Manual for General, Organic, and Biochemistry** Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives Essentials of General Organic and Biochemistry captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit [www.whfreeman.com/gob](http://www.whfreeman.com/gob) **Chemistry** Karen Timberlake, 1999 Suitable for one or two term lab courses covering general organic and biological chemistry this new edition written by Karen Timberlake features many improvements to the insightful experiments that have made it the leading lab manual Each experiment encourages critical thinking with laboratory goals discussion of related concepts clear instructions new pre lab questions and comprehensive report pages Forty one experiments illustrate the basic principles of chemistry **The Essential Lab Manual** Karen Timberlake, 2002-06-24 Drawing from the successful main Laboratory Manual the Essential Laboratory Manual includes twenty one experiments which have been revised and updated Suitable for a one or two term lab course **Report**, 1924

**General, Organic, and Biological Chemistry Study Guide and Selected Solutions** Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems **Surfactants in**

**Solution** K.L. Mittal, P. Botherel, 2012-12-06 This and its companion Volumes 4 and 5 document the proceedings of the 5th International Symposium on Surfactants in Solution held in Bordeaux France July 9-13 1984 This symposium was the continuation of the series of symposia initiated in 1976 in Albany New York under the title icellization Solubilization and icroemulsions The next two symposia were labelled Solution Chemistry of Surfactants and Solution Behavior of Surfactants Theoretical and Applied Aspects held in Knoxville TN in 1978 and Potsdam N Y in 1980 respectively In 1982 at the time of

the 4th Symposium in this series it became amply evident that there was a definite need to have more a generic title to describe these biennial events and after much deliberation it was decided that an appropriate title would be Surfactants in Solution as both the aggregation and adsorption aspects of surfactants were addressed So the 4th Symposium was held in 1982 in Lund Sweden under this new rubric and it was decided to continue these symposia in the future under this appellation Naturally the Bordeaux Symposium was dubbed as the 5th International Symposium on Surfactants in Solution and our logo became SIS which is very apropos and appealing It was in Bordeaux that the decision was made to hold the 6th SIS Symposium in New Delhi and it is scheduled for August 18 22 1986 in the capital of India Prentice Hall Science

Explorer: Chemical Interactions Pearson/Prentice Hall, Michael J. Padilla, 2004-10 **U.S. Government Research Reports**, 1964 **Scientific and Technical Aerospace Reports**, 1995 **Energy Research Abstracts**, 1989 **Applied**

**Mechanics Reviews**, 1985 Oswaal ISC Question Bank Class 12 Chemistry| Chapterwise and Topicwise | Solved Papers |

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