

EVALUATE: The conversion $1 \text{ y} = 3.156 \times 10^7 \text{ s}$ assumes $1 \text{ y} = 365.24 \text{ d}$, which is the average for one extra day every four years, in leap years. The problem says instead to assume a 365-day year.

1.8. IDENTIFY: Apply the given conversion factors.

SET UP: 1 furlong = 0.1250 mi and 1 fortnight = 14 days. 1 day = 24 h.

EXECUTE: $(180,000 \text{ furlongs/fortnight}) \left(\frac{0.125 \text{ mi}}{1 \text{ furlong}} \right) \left(\frac{1 \text{ fortnight}}{14 \text{ days}} \right) \left(\frac{1 \text{ day}}{24 \text{ h}} \right) = 67 \text{ mi/h}$

EVALUATE: A furlong is less than a mile and a fortnight is many hours, so the speed limit in mph is a much smaller number.

1.9. IDENTIFY: Convert miles/gallon to km/L.

SET UP: 1 mi = 1.609 km. 1 gallon = 3.788 L.

EXECUTE: (a) $55.0 \text{ miles/gallon} = (55.0 \text{ miles/gallon}) \left(\frac{1.609 \text{ km}}{1 \text{ mi}} \right) \left(\frac{1 \text{ gallon}}{3.788 \text{ L}} \right) = 23.4 \text{ km/L}.$

(b) The volume of gas required is $\frac{1500 \text{ km}}{23.4 \text{ km/L}} = 64.1 \text{ L}.$ $\frac{64.1 \text{ L}}{45 \text{ L/tank}} = 1.4 \text{ tanks}.$

EVALUATE: 1 mi/gal = 0.425 km/L. A km is very roughly half a mile and there are roughly 4 liters in a gallon, so 1 mi/gal = $\frac{1}{2}$ km/L, which is roughly our result.

1.10. IDENTIFY: Convert units.

SET UP: Use the unit conversions given in the problem. Also, 100 cm = 1 m and 1000 g = 1 kg.

EXECUTE: (a) $\left(60 \frac{\text{mi}}{\text{h}} \right) \left(\frac{1 \text{ h}}{3600 \text{ s}} \right) \left(\frac{5280 \text{ ft}}{1 \text{ mi}} \right) = 88 \frac{\text{ft}}{\text{s}}$

(b) $\left(32 \frac{\text{ft}}{\text{s}^2} \right) \left(\frac{30.48 \text{ cm}}{1 \text{ ft}} \right) \left(\frac{1 \text{ m}}{100 \text{ cm}} \right) = 9.8 \frac{\text{m}}{\text{s}^2}$

(c) $\left(1.0 \frac{\text{g}}{\text{cm}^3} \right) \left(\frac{100 \text{ cm}}{1 \text{ m}} \right)^3 \left(\frac{1 \text{ kg}}{1000 \text{ g}} \right) = 10^3 \frac{\text{kg}}{\text{m}^3}$

EVALUATE: The relations $60 \text{ mi/h} = 88 \text{ ft/s}$ and $1 \text{ g/cm}^3 = 10^3 \text{ kg/m}^3$ are exact. The relation $32 \text{ ft/s}^2 = 9.8 \text{ m/s}^2$ is accurate to only two significant figures.

1.11. IDENTIFY: We know the density and mass; thus we can find the volume using the relation

density = mass/volume = m/V . The radius is then found from the volume equation for a sphere and the result for the volume.

SET UP: Density = 19.5 g/cm^3 and $m_{\text{sphere}} = 60.0 \text{ kg}$. For a sphere $V = \frac{4}{3}\pi r^3$.

EXECUTE: $V = m_{\text{sphere}}/\text{density} = \left(\frac{60.0 \text{ kg}}{19.5 \text{ g/cm}^3} \right) \left(\frac{1000 \text{ g}}{1.0 \text{ kg}} \right) = 3080 \text{ cm}^3.$

$$r = \sqrt[3]{\frac{3V}{4\pi}} = \sqrt[3]{\frac{3}{4\pi}(3080 \text{ cm}^3)} = 9.0 \text{ cm}.$$

EVALUATE: The density is very large, so the 130 pound sphere is small in size.

1.12. IDENTIFY: Use your calculator to display $\pi \times 10^7$. Compare that number to the number of seconds in a year.

SET UP: 1 yr = 365.24 days. 1 day = 24 h, and 1 h = 3600 s.

EXECUTE: $(365.24 \text{ days/1 yr}) \left(\frac{24 \text{ h}}{1 \text{ day}} \right) \left(\frac{3600 \text{ s}}{1 \text{ h}} \right) = 3.15567... \times 10^7 \text{ s};$ $\pi \times 10^7 \text{ s} = 3.14159... \times 10^7 \text{ s}$

The approximate expression is accurate to two significant figures.

EVALUATE: The close agreement is a numerical accident.

1.13. IDENTIFY: The percent error is the error divided by the quantity.

SET UP: The distance from Berlin to Paris is given to the nearest 10 km.

EXECUTE: (a) $\frac{10 \text{ m}}{890 \times 10^3 \text{ m}} = 1.1 \times 10^{-5}\%$.

(b) Since the distance was given as 890 km, the total distance should be 890,000 meters. We know the total distance to only three significant figures.

EVALUATE: In this case a very small percentage error has disastrous consequences.

1.14. IDENTIFY: When numbers are multiplied or divided, the number of significant figures in the result can be no greater than in the factor with the fewest significant figures. When we add or subtract numbers it is the location of the decimal that matters.

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