



Women's Compulsory Program

USA Gymnastics Women's Compulsory Program

Level 1

Level 2

Level 3

Level 4

Level 5

Conditioning

Flexibility

Penalties

Glossary



USA GYMNASTICS.

Usa Gymnastics Compusory Routines

Roman Wölfel

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arrhenius model of bases a substance that contains a hydroxide group and dissociates to produce a hydroxide

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web kw the product of the concentrations of the hydrogen ions and hydroxide ions acidic solution one which h is greater than

the oh basic solution one in which the h is less than the oh alkaline solutions the h of a basic solution is less than 1×10^{-7} m

they are the same as basic solutions

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nh₂ and s²⁻ are ch₄ nh₃ and hs⁻ respectively next we consider that acidity increases with positive charge on the molecule

thus ruling out that s²⁻ is the weakest base

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molecule that gains a hydrogen ion becomes this which is positively charged what did lewis propose an acid accepts a pair of

electrons during a reaction while a base donates a pair of electrons

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