

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____

Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	152	Sodium - 11	190	Potassium - 19	243	Rubidium - 27	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 28	219
Boron - 5	87	Aluminum - 13	118	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 52	125	Tin - 50	145
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 27	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 28	550

Atomic Radius



Trends In The Periodic Table Answers Graphing

JE Gale



Trends In The Periodic Table Answers Graphing:

Chemistry in the Community (Enhanced Core Four) American Chemical Society, 2006-02-15 Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. WileyPLUS sold separately from text.

Chemistry in the Community. American Chemical Society, 2002. This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts.

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M. Merrill,2010-09-15 This workbook was written for students of epidemiology and serves as a supplement to any one of several introductory text books in epidemiology Each chapter is divided into an introduction a series of questions and detailed responses and a series of Homework questions At the end of each chapter is a table with a list of selected epidemiology text books with accompanying chapters in those books that the workbook chapter may supplement The general learning outcomes LOs for this workbook are 1 Become familiar with basic concepts and definitions commonly used in epidemiology 2 Define a public health problem 3 Identify appropriate uses and limitations of data and research design strategies for solving public health problems 4 Make relevant inferences from quantitative and qualitative data 5 Distinguish between statistical association and cause effect relationships 6 Measure and describe patterns of disease incidence prevalence and mortality 7 Identify environmental factors and behaviors associated with health related states or events 8 Be familiar with the steps for investigating disease outbreaks 9 Identify calculate and interpret common indices used in identifying the health status 10 Evaluate program effectiveness 11 Critically assess epidemiological research 12 Be able to communicate health findings Each chapter features 10 20 mastery check questions with detailed answers 5 optional problems A case study A multiple choice short answer quiz Answers to the cases and quizzes are provided as part of the online instructor resource package Hands-On General Science Activities With Real-Life Applications Pam Walker,Elaine Wood,2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life *Chemistry Homework for OCR A for Double and Separate Awards* Gareth Pritchard,2001 Howework activites for OCR A Chemistry specifications **Understanding Advanced Physical Inorganic Chemistry: The Learner's Approach (Revised Edition)** Kim Seng Chan,Jeanne Tan,2016-09-26 This revised edition has been updated to meet the minimum requirements of the new Singapore GCE A level syllabus that would be implemented in the year 2016 Nevertheless this book is also highly relevant to students who are studying chemistry for other examination boards In addition the authors have also included more Q A to help students better understand and appreciate the chemical concepts that they are mastering *NEST : National Entrance Screening Test (General Aptitude, Physics, Chemistry, Mathematics, Biology) - 10 Mock Tests (Solved Questions) with Free Access to Online Tests* EduGorilla Prep Experts,2020-12-28 National Entrance Screening Test NEST is conducted jointly by National Institute of Science Education and Research NISER Bhubaneswar and Mumbai University The candidates that are able to successfully qualify the exams will be offered admission in the above mentioned prestigious universities The NEST Exam is conducted annually to offer admission for Integrated MSc programme in Biology Chemistry Mathematics Physics *CBSE Most Likely Question Bank Science Class 10 (2022 Exam) - Categorywise & Chapterwise with New Objective Paper Pattern, Reduced Syllabus*

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