VSEPR Theory Chart

Number of Electron Groups	Lone Pairs = 0	Lone Pairs = 1	Lone Pairs = 2	Lone Pairs = 3	Lone Pairs = 4
2	Linear				
3	Trigonal Planar	Angular or Bent			
4	Tetrahedral	Trigonal Pyramidal	Angular or Bent		
5	Trigonal Bipyramidal	Seesaw	T-shaped	Linear	
6	Octahedral	Square Pyramidal	Square Planar	T-shaped	Linear

Vsepr Theory Test

Ashima Arora, Shahbaz, NK Yadav, Anand Singh, Poonam Dutta, AK Seejo, Anshika Verma, Pallavi Kaushik, Raju Singh, Kalpna Seejo, Swati Saluja

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