

**Chapter
Review****Chemical Reactions****Part A. Vocabulary Review**

Directions: Complete the following sentences using the terms listed below.

synthesis **exothermic** **catalyst** **decomposition** **coefficients**
endothermic **single-displacement** **inhibitor** **reactants**
precipitate **chemical** **double-displacement** **products** **balanced**

1. A change in which one or more substances are converted into a new substance is called a(n) **double-displacement** reaction.

2. If a precipitate, water, or a gas forms when two ionic compounds in solution are combined, the reaction is a(n) **precipitate** reaction.

3. A substance that prevents certain chemical reactions is a(n) **inhibitor**.

4. When two or more substances combine in a chemical reaction to form another substance, a(n) **synthesis** reaction has taken place.

5. The breakdown of a substance into two or more simpler substances is a(n) **decomposition** reaction.

6. In a chemical equation, the substances that react are called **reactants**.

7. A substance that speeds up a chemical reaction without itself being permanently changed is a(n) **catalyst**.

8. A reaction that requires energy in the form of heat is a(n) **endothermic** reaction.

9. When one element replaces another in a compound, a(n) **single-displacement** reaction has occurred.

10. If the energy released in a chemical reaction is primarily in the form of heat, that reaction is a(n) **exothermic** reaction.

11. Numbers in front of the symbols and formulas in an equation are called **coefficients**.

12. If the number of atoms of each element on the left side of an equation is equal to the number of atoms of each element on the right side of the equation, the equation is **balanced**.

13. In a chemical equation, the substances that are produced are called **products**.

14. An insoluble solid that forms when two ionic compounds in solution are combined is a(n) **chemical** reaction.

Vocabulary Review Answer Key Chemistry

**Melanie Williams, Herbert
Puchta, Günter Gerngross, Peter Lewis-
Jones**

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Table of Contents Vocabulary Review Answer Key Chemistry

1. Understanding the eBook Vocabulary Review Answer Key Chemistry
 - The Rise of Digital Reading Vocabulary Review Answer Key Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Vocabulary Review Answer Key Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Vocabulary Review Answer Key Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Vocabulary Review Answer Key Chemistry
 - Personalized Recommendations
 - Vocabulary Review Answer Key Chemistry User Reviews and Ratings

- Vocabulary Review Answer Key Chemistry and Bestseller Lists
- 5. Accessing Vocabulary Review Answer Key Chemistry Free and Paid eBooks
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 - Adjustable Fonts and Text Sizes of Vocabulary Review Answer Key Chemistry
 - Highlighting and Note-Taking Vocabulary Review Answer Key Chemistry
 - Interactive Elements Vocabulary Review Answer Key Chemistry
- 8. Staying Engaged with Vocabulary Review Answer Key Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Vocabulary Review Answer Key Chemistry
- 9. Balancing eBooks and Physical Books Vocabulary Review Answer Key Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Vocabulary Review Answer Key Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Vocabulary Review Answer Key Chemistry
 - Setting Reading Goals Vocabulary Review Answer Key Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Vocabulary Review Answer Key Chemistry
 - Fact-Checking eBook Content of Vocabulary Review Answer Key Chemistry
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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