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Chapter 3: Thermodynamic Principles: A Review

Matching

- A) q/T
- B) entropy
- C) $H - TS$
- D) equilibrium
- E) ΔG°

1. The term used to indicate the degree of randomness within a system is _____.

Ans: B

Level of Difficulty: Easy

Section: 3-2, Second Law of Thermodynamics: The Universe Tends Toward Maximum Disorder

2. For spontaneous processes within biological systems at constant temperature, $\Delta S \geq$ _____.

Ans: A

Level of Difficulty: Easy

Section: 3-2, Second Law of Thermodynamics: The Universe Tends Toward Maximum Disorder

3. Gibbs free energy is defined as $G =$ _____.

Ans: C

Level of Difficulty: Easy

Section: 3-3, Free Energy: The Indicator of Spontaneity

4. The symbol for free energy under standard biochemical conditions is _____.

Ans: E

Level of Difficulty: Easy

Section: 3-3, Free Energy: The Indicator of Spontaneity

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Biology ,1971 *Index Veterinarius* ,1998 *The New York Times Index* ,1998 *Proceedings of the National Academy of Sciences of the United States of America* National Academy of Sciences (U.S.),1986 Biochemistry Donald Voet,Judith G. Voet,1990-01-16 This comprehensive introductory text thoroughly explains basic biochemical concepts while offering a unified presentation of the field and its development Emphasizes biochemistry as a body of knowledge compiled through experimentation stresses the unity of life and its variation through evolution and the ways in which biological processes are organized into interdependent networks Also examines medical applications of biochemical knowledge identifying some of the major contributors to the field and approaches they have taken to solve important biochemical puzzles Contains excellent art carefully planned for pedagogical impact including illustrations by Irving Geis Current extensive references and creative problem sets are also included **Who's who in the East** ,1985

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