

Final Activity Series

(This is similar to the "Activity Series" you learned in class.)

Notes:

These tables provide experimental results to give you information to determine activity series. They require that you know the relative activity of metals. A metal is more active than a metal ion if it can displace that metal from its ionic compound. Which metals can react, including potassium, nickel and metals forming metal. This activity series is not complete, but it is a good starting point.

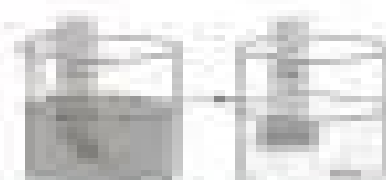
Model 1 – Reaction of Metals



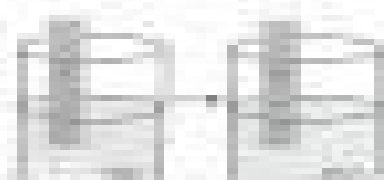
Reaction 1: $M + M^+ \rightarrow M + M^+$



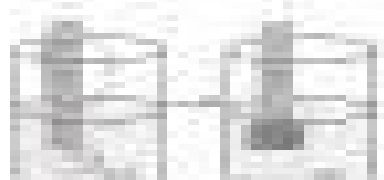
Reaction 2: $M + M^+ \rightarrow M + M^+$



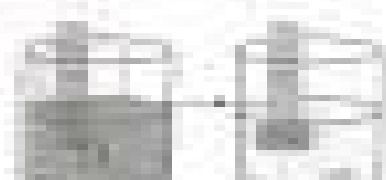
Reaction 3: $M + M^+ \rightarrow M + M^+$



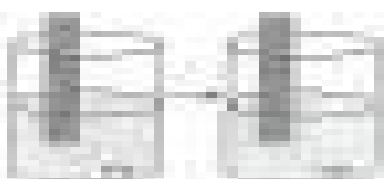
Reaction 4: $M + M^+ \rightarrow M + M^+$



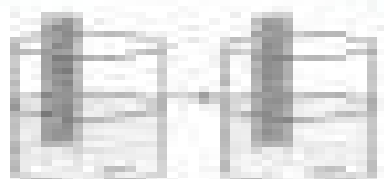
Reaction 5: $M + M^+ \rightarrow M + M^+$



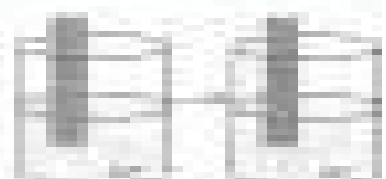
Reaction 6: $M + M^+ \rightarrow M + M^+$



Reaction 7: $M + M^+ \rightarrow M + M^+$



Reaction 8: $M + M^+ \rightarrow M + M^+$



Reaction 9: $M + M^+ \rightarrow M + M^+$

Notes: These are experimental results, not theoretical. They are based on experimental data.

The Activity Series Pogil

Sally A. Fincher, Anthony V. Robins



The Activity Series Pogil:

The Activity Series Pogil Book Review: Unveiling the Power of Words

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