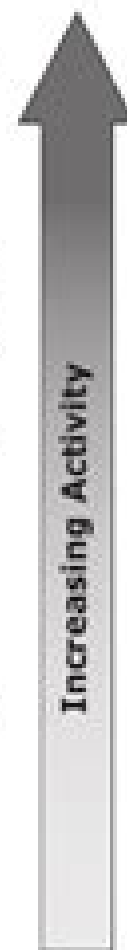


# STAAR CHEMISTRY REFERENCE MATERIALS



POLYATOMIC IONS		SOLUBILITY OF COMMON IONIC COMPOUNDS IN WATER		ACTIVITY SERIES
Acetate	$\text{C}_2\text{H}_3\text{O}_2^-$ , $\text{CH}_3\text{COO}^-$	<u>Soluble</u> <u>compounds contain</u>	<u>Common exceptions</u>	<u>Metal</u>
Ammonium	$\text{NH}_4^+$	$\text{C}_2\text{H}_3\text{O}_2^-$ , $\text{CH}_3\text{COO}^-$	None	Lithium
Carbonate	$\text{CO}_3^{2-}$	$\text{NH}_4^+$	None	Potassium
Chlorate	$\text{ClO}_3^-$	$\text{NO}_3^-$	None	Barium
Chlorite	$\text{ClO}_2^-$	$\text{CN}^-$	None	Calcium
Chromate	$\text{CrO}_4^{2-}$	$\text{ClO}^-$	None	Sodium
Cyanide	$\text{CN}^-$	$\text{CO}_3^{2-}$	None	Magnesium
Dichromate	$\text{Cr}_2\text{O}_7^{2-}$	$\text{ClO}_2^-$	None	Aluminum
Hydrogen carbonate	$\text{HCO}_3^-$	$\text{ClO}_3^-$	None	Manganese
Hydroxide	$\text{OH}^-$	$\text{Br}^-$	Compounds of $\text{Ag}^+$ , $\text{Pb}^{2+}$ , and $\text{Hg}_2^{2+}$	Zinc
Hypochlorite	$\text{ClO}^-$	$\text{Cl}^-$	Compounds of $\text{Ag}^+$ , $\text{Pb}^{2+}$ , and $\text{Hg}_2^{2+}$	Chromium
Nitrate	$\text{NO}_3^-$	$\text{I}^-$	Compounds of $\text{Ag}^+$ , $\text{Pb}^{2+}$ , and $\text{Hg}_2^{2+}$	Iron
Nitrite	$\text{NO}_2^-$	$\text{SO}_4^{2-}$	Compounds of $\text{Sr}^{2+}$ , $\text{Ba}^{2+}$ , $\text{Pb}^{2+}$ , and $\text{Hg}_2^{2+}$	Cobalt
Perchlorate	$\text{ClO}_4^-$	<u>Insoluble</u> <u>compounds contain</u>	<u>Common exceptions</u>	Nickel
Permanganate	$\text{MnO}_4^-$	$\text{CO}_3^{2-}$	Compounds of $\text{NH}_4^+$ and the alkali metal cations	Tin
Phosphate	$\text{PO}_4^{3-}$	$\text{PO}_4^{3-}$	Compounds of $\text{NH}_4^+$ and the alkali metal cations	Lead
Sulfate	$\text{SO}_4^{2-}$	$\text{CrO}_4^{2-}$	Compounds of $\text{NH}_4^+$ and the alkali metal cations	(Hydrogen)
Sulfite	$\text{SO}_3^{2-}$	$\text{Cr}_2\text{O}_7^{2-}$	Compounds of $\text{NH}_4^+$ and the alkali metal cations	Copper
		$\text{OH}^-$	Compounds of $\text{NH}_4^+$ , the alkali metal cations, $\text{Ca}^{2+}$ , $\text{Sr}^{2+}$ , and $\text{Ba}^{2+}$	Mercury
		$\text{S}^{2-}$	Compounds of $\text{NH}_4^+$ , the alkali metal cations, $\text{Ca}^{2+}$ , $\text{Sr}^{2+}$ , and $\text{Ba}^{2+}$	Silver
				Platinum
				Gold



# Staar Reference Material Chemistry Activities

**S Ben Porath**



## **Staar Reference Material Chemistry Activities:**

## Unveiling the Magic of Words: A Report on "**Staar Reference Material Chemistry Activities**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Staar Reference Material Chemistry Activities**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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