

- (ii) State the Raoult's law of relative lowering of vapour pressure of a solution. 1 + 1

OR

- What is osmotic pressure? 2
- (b) Write two differences between physical adsorption and chemical adsorption. 2

OR

What is emulsion? Give an example of water in oil type emulsion. 1 + 1

- (c) (i) Why does  $\text{PCl}_3$  produce white fumes in moist air? 2
- (ii) Why is  $\text{ICl}$  more reactive than  $\text{I}_2$ ? 1 + 1

OR

Deduce the molecular shape of  $\text{BrF}_3$  on the basis of VSEPR theory. 2

- (d) (i) Calculate the oxidation number of central metal atom of the following complex compound :



- (ii) When aqueous solution of one mole of  $\text{CrCl}_3 \cdot 6\text{H}_2\text{O}$  complex compound is treated with excess of  $\text{AgNO}_3$ , 3 moles of  $\text{AgCl}$  are precipitated. Determine the formula of the complex compound. 1 + 1

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**A Gutmann**



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