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Quiz: Types of Chemical Reactions True/False (1 point each)

- In a decomposition reaction, a single compound breaks down into two or more simpler substances.
- Combustion reactions always involve oxygen as a reactant.
- A chemical reaction that releases energy is called an endothermic reaction.
- In a double-replacement reaction, the positive and negative lons of two reactants switch places.
- The reactants in a chemical equation are found on the right side of the arrow.

Multiple Choice (1 point each)

- 6. Which of the following is the general form for a synthesis reaction? a) AB → A + B b) A + B → AB c) A + BC → AC + B d) AB + CD → AD + CB
- What type of reaction is represented by the following equation: 2H₂O(I) → 2H₂(g) + O₂(g)? a) Synthesis b)
 Decomposition c) Single-replacement d) Double-replacement
- B. in a single-replacement reaction, a more reactive element replaces alless reactive element in a compound. Which of the following metals is the MOST reactive? a) Copper (Cu) b) Zinc (Zn) c) Silver (Ag) d) Gold (Au)
- Which of the following is an example of a combustion reaction? a) 2Nals) + Cl₂(g) → 2NaCl(s) b) CaCO₂(s) → CaO(s) + CO₂(g) c) CH₂(g) + 2O₂(g) → CO₂(g) + 2H₂O(l) d) Fe(s) + CuSO₂(ag) → FeSO₂(ag) + Cu(s)
- What type of reaction is represented by the following equation: AgNOy(sq) + NaCl(sq) → AgCl(s) + NaNOy(sq)? a) Synthesis b) Decomposition c) Single-replacement d) Doublereplacement
- Which of the following is NOT a sign that a chemical reaction has occurred? a) Change in color b) Formation of a precipitate c) Change in state d) Production of gas
- What is the name for a solid that forms from a solution during a chemical reaction? a) Catalyst b) Precipitate c) Reactant d) Product

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