Street,

PAPER-2 [Code - 5] IITJEE 2011 PART - IR CHEMISTR

SECTION - I (Total Marks : 24) (Single Correct Answer Type)

This Section committee 8 multiple choice questions. Each question has fine choices (A), (B), (C) and (D) out of which ONLY ONLy contrar.

- The freezing point (in "C) of a solution containing this g of K_(FactCN)₀ (Mot. Wt. 329) in 100 g of water (K₁ = 1.56 K kg small⁻¹) is
 A T = 1.07
 - $(A) 2.3 \times 10^{-3}$ $(B) 5.7 \times 10^{-3}$ $(C) 5.7 \times 10^{-3}$ $(C) 1.2 \times 10^{-3}$
- Not. (A): $K_{-}[fw(CN)]_{L}] \rightarrow 3K^{+} + [fw(CN)]_{L}]^{+}$ 1 = 4 $\Delta T_{-} = K_{+} \times 1 \times \frac{m}{M} \times \frac{1000}{W} = 1.86 \times 4 \times \frac{0.1}{329} \times \frac{1000}{100} = 2.3 \times 10^{-1}$ $T_{+} = -2.3 \times 10^{-1}$
- Assungs) the compounds given, the one that would form a brilliant colored dye on restment with NaNO, in dil. BCI followed by addition to an alkaline solution of \$5 naphthed is

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A Gutmann

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