Molecular Structure

Instructions: Fill in each section of this worksheet and submit it to your instructor for grading.

Review Questions

- Write a statement explaining which characteristics of an atom determine the VSEPR shape of a molecule (hint: look at the VSEPR geometry chart).
 VSEPR shows the shape of a molecule and the arrangement of te electron pairs and the atom, it predicts the lone pairs and it predicts shape.
- How does the force between the atoms impact the shape of a molecule? The force between the atom determines the shape.

Activity: First row is done for you as a model.

Compound	Draw Lewis Structure	Use VSEPR Chart to determine:	Determine the structures of the molecule	Draw the VSEPR model
H ₂ O	н- ё- н	Total Domains: 4 Bonded Atoms: 2 Lone Pairs:	Molecular Shape: Bent Electron Geometry: Tetrahedral	Ю. Н
SiH.	H-0-H	Total Domains: 4 Bonded Atoms: 4 Lone Pairs: 0	Molecular Shape: tetrahedral Electron Geometry: tetrahedral	H.S. H.
Br ₃	B-B	Total Domains: 1 Bonded Atoms: 1 Lone Pairs: 0	Molecular Shape: linear Electron Geometry: linear	B = B

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South Pasedena Bonding And Molecular Structure Worksheet Answers

JA Banks

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