

**Q 3. Answer the following questions.**

- (i) Discuss law of Reciprocal Proportion with examples.
- (ii) An inorganic salt gave the following percentage composition:  
Na = 29.11%, S = 40.51% & O = 30.38%, Calculate the Empirical Formula of the salt.
- (iii) Discuss Aufbau's Principle and Hund's Rule.
- (iv) What is diagonal relationship with example.
- (v) Explain  $sp^2$  hybridization with example.
- (vi) On the basis of VSEPR theory, predict the shapes of the following molecules/ions?  
(a)  $SF_4$  (b)  $NH_4^+$  (c)  $H_3O^+$   
(d)  $NH_3$  (e)  $PCl_3$  (f)  $NH_2$
- (vii) State & explain Hess's Law of constant heat summation.
- (viii) What is Buffer Solution. Give its types.
- (ix) Give three methods of preparation of Alkynes.

## SECTION D

**Q 4. Answer the following questions.**

- (i) What are Quantum numbers? Explain  
Or  
State and Explain Heisenberg 's Uncertainty Principle.
- (ii) Define 1st Law of Thermodynamics. Derive its mathematical formulation.  
Or  
Define Heat Capacity. Derive relation between  $C_p$  &  $C_v$ .
- (iii) Explain Nucleophilic Substitution Reaction.  
Or  
Write the IUPAC names of the following organic compounds:

- (a)  $\text{CH}_3 - \text{CH}_2 - \text{CH} = \text{CH} - \text{CHO}$   
(b)  $\begin{array}{c} \text{COOH} \\ | \\ \text{COOH} \end{array}$

*[Signature]*

# Vbcat Chemistry Question Paper

**Ajnish Kumar Gupta, Bharti Gupta**

A red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the authors' names.

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