

8 Structure and nomenclature of organic compounds

Unit	4	How are organic compounds categorised, analysed and used?
Area of Study	1	How can the diversity of carbon compounds be explained and categorised?
Topic	8	Structure and nomenclature of organic compounds
Subtopic	8.2	The carbon atom

Summary

- Carbon atoms have 4 valence electrons and make 4 covalent bonds with other carbon atoms or atoms of different elements.
- Bond energy is the amount of energy required to break the bonds of reactants or the amount of energy released per mole when a covalent bond forms between 2 molecules. It is typically measure in kJ mol^{-1} .
- The bond length is the distance between the nuclei of the atoms sharing the electrons.
- Carbon molecules with single bonds typically form a tetrahedral arrangement.
- Double $\text{C}=\text{C}$ bonds are stronger than $\text{C}-\text{C}$ single bonds and forms flat molecules. $\text{C}\equiv\text{C}$ bonds are stronger again and form linear molecules.



My notes

Vcaa Chemistry Study Guide

Richard Morris



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The early part of the book explicates the transdisciplinary negotiated theoretical underpinning of the MDL framework followed by the research informed repertoire of learning experiences that are then articulated into a comprehensive framework of options for the planning of classroom work

Practical adoption and adaptation of the framework in biology chemistry and physics classrooms are detailed in separate chapters The latter chapters indicate the impact of the collaborative research on teachers professional learning and students multimodal disciplinary literacy engagement

concluding with proposals for accommodating emerging developments in MDL in an ever changing digital communication world The MDL framework is designed to enable teachers to develop all students disciplinary literacy competencies

This book will be of interest to researchers teacher educators and postgraduate students in the field of science education It will also have appeal to those in literacy education and social semiotics

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