

Experiment 2 - Gravimetric Analysis of a Chloride Salt (G)

Purpose:

Determining quantitative value of chloride in an unknown soluble salt by showing and explaining common methods in gravimetric analysis.

Theory:

- Reaction between silver ion and chloride ion to form silver chloride as an equation:



- AgCl solubility in water and the K_{sp} :

Silver chloride is very insoluble, so we can assume that the above reaction goes to completion.

$$K_{sp} = 1.6 \times 10^{-10}$$

- AgCl is precipitated with an acidic solution to reduce likelihood of acid ions interfering with results. These ions will form co-precipitates with silver in the acid, due to these co-precipitates, the results would be higher.
- To prevent the precipitate from going through the filtered paper, it is heated to coagulate it into a different texture that's easier to work with. Heating the precipitate makes it coagulate, therefore, it is harder for the precipitate to go through the filtered paper.
- Dry silver chloride decomposes into silver and chlorine in the presence of light:



- If photodecomposition takes place in air, results will be lower, since chlorine will be released in the form of gas. If photochemical decomposition occurs in the presence of excess silver ions in aqueous solution, it will result in higher results.

Unknown Salt Lab Report

Harold R. Hunt, Toby F. Block



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English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента Надежда Зорина, Александра Соболева, 2022-10-19 *Lab Manual for Investigating Chemistry* David Collins (Ph. D.), Matthew

Johll, 2008-12-02 While many of the core labs from the first edition have been retained a renewed focus on the basics of chemistry and the scientific process create an even more detailed supplemental offering **Modern Experimental**

Chemistry George W. Jr. Latimer, 2012-12-02 Modern Experimental Chemistry provides techniques of qualitative analysis that reinforce experiments on ionic equilibria This book includes the determination of water in hydrated salts identification of an organic compound after determining its molecular weight and nonaqueous titration of a salt of a weak acid The calculation of chemical stoichiometry calculation of thermodynamic properties by determining the change in equilibrium with temperature and chromium chemistry are also covered This compilation contains enough experiments for classes which have six hours of laboratory two 3 hour meetings per week to last two semesters This publication is intended for chemistry students as an introductory manual to chemistry laboratory **The Essential Lab Manual** Karen

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techniques to coax out the reluctant scientists in your classroom This book is full of classroom tested pragmatic approaches from high school science teachers who used the ideas to make teaching and learning more creative endeavors *Wren's Story* Wren Richards,2020-07-30 Wren was abandoned into the Wyoming wilderness to grow with an education based in knowledge from nature and the Indian Mission Abandonment taught her to use the environment as her teacher The isolated Wind River Reservation culture taught her resourcefulness individuality self sufficiency and durability The laws from nature promoted reflectiveness and meditation where Sun Wren Richards learned that all questions were answered with transcendence from within Subject Index to Unclassified ASTIA Documents Defense Documentation Center (U.S.),1960

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