

8 Structure and nomenclature of organic compounds

Unit	4	How are organic compounds categorised, analysed and used?
Area of Study	1	How can the diversity of carbon compounds be explained and categorised?
Topic	8	Structure and nomenclature of organic compounds
Subtopic	8.2	The carbon atom

Summary

- Carbon atoms have 4 valence electrons and make 4 covalent bonds with other carbon atoms or atoms of different elements.
- Bond energy is the amount of energy required to break the bonds of reactants or the amount of energy released per mole when a covalent bond forms between 2 molecules. It is typically measure in kJ mol^{-1} .
- The bond length is the distance between the nuclei of the atoms sharing the electrons.
- Carbon molecules with single bonds typically form a tetrahedral arrangement.
- Double $\text{C}=\text{C}$ bonds are stronger than $\text{C}-\text{C}$ single bonds and forms flat molecules. $\text{C}\equiv\text{C}$ bonds are stronger again and form linear molecules.



My notes

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**Steven Morris, Michael Sheedy, James
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disciplinary literacy MDL fused with the subject specific science pedagogies of senior high school biology chemistry and physics It builds a compatible alignment of multiple representation and representation construction approaches to science pedagogy with the social semiotic systemic functional linguistic based approaches to explicit teaching of disciplinary literacy

The early part of the book explicates the transdisciplinary negotiated theoretical underpinning of the MDL framework

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framework of options for the planning of classroom work Practical adoption and adaptation of the framework in biology

chemistry and physics classrooms are detailed in separate chapters The latter chapters indicate the impact of the

collaborative research on teachers professional learning and students multimodal disciplinary literacy engagement

concluding with proposals for accommodating emerging developments in MDL in an ever changing digital communication

world The MDL framework is designed to enable teachers to develop all students disciplinary literacy competencies This

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