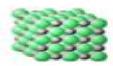
#### Unit 5 Covalent Bonding Webquest

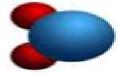
#### Activity #1- Introduction to Covalent Bonding

Open Chemical Bonding http://www.visionlearning.com/en/library/Chemistry/1/Chemical-Bonding/55 Scroll down to the heading "Covalent Bonding."

As opposed to	bonding in which a
complete transfer of electrons o	
bonding occurs when two or mor	re elements
electrons. Covalent bonding occ	urs because the atoms in the
compound have a similar tenden	cy for electrons (generally to
electr	rons). This most commonly occurs
when two	bond together. Because both of
the nonmetals will want to	electrons, the
elements involved will share elec-	ctrons in an effort to
their valence	shells.

- 2. How many valence electrons are in one atom of hydrogen?
- 3.How many valence electrons does hydrogen need to have a full first shell?
- 4. How does the hydrogen atom "pick up" another electron?
- 5. What compound does hydrogen form?
- 6. How do hydrogen atoms make a covalent bond?
- Label the substances below as "ionic" or "covalent"





# **Webquest Answer Key For Unit 4 Covalent Bonding**

**Ying-Ying Zheng** 

Webquest Answer Key For Unit 4 Covalent Bonding:

## Reviewing Webquest Answer Key For Unit 4 Covalent Bonding: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "Webquest Answer Key For Unit 4 Covalent Bonding," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

http://www.frostbox.com/About/virtual-library/default.aspx/split\_back\_veer\_offense\_playbook.pdf

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