

MKC STUDY GUIDE

REVERSE OSMOSIS (MK3)

- **Theory:** Pressurized seawater is passed over a semi permeable membrane that passes pure water but excludes salt water.
- Osmotic pressure is the difference in pressure across the membrane.
- Reverse Osmosis is known as Hyper-filtration.
- Osmotic pressure is equal to 1 psi per 100 milligrams per liter of salt.
- The osmotic pressure of sea water is **350-370 psi**.
- Osmotic pressure occurs when pressure is applied to the higher salt concentration side of the membrane, causing pure water to be forced out of the salty solution.
- **Low Pressure Switch:** Secures the high press pump if the low press switch pre-determined setting isn't met (**on 110's setting is 5 psi**).
- **High Pressure Pump:** Creates a pressure of **700-1000 psi** to achieved reverse osmosis.
- **Pulsation Dampener:** Absorbs pulsations created on discharge side of high press pump. Consists of a nitrogen charged rubber bladder charged to approximately **2/3 system pressure**.
- Cracking pressure for system relief valve is most likely set at **1050 psi**.
- **Pressure Regulator:** Normally a needle valve installed on the brine discharge line used to keep the module pressure at **700-1000 psi**.
- When the salinity cell detects more than **500 ppm tds (salt)** it shifts the 3 way valve to overboard discharge position. **Audio & visual alarm notification**.
- **Crossflow:** Allows the membrane to continually clean itself. Fluid passes through the membrane while the rest continues downstream, sweeping the rejected species away from the membrane.
- **Daltons:** Atomic Mass Unit (measure of atom's mass). A unit of mass defined as 1/12 of the mass of a carbon-12 atom, equal to 1.66×10^{-27} kg.
- **John Dalton:** British physicist and chemist that developed the atomic theory of elements and molecules, founder of modern physical science.

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JL Elias

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Table of Contents Uscg Mkc Study Guide

1. Understanding the eBook Uscg Mkc Study Guide
 - The Rise of Digital Reading Uscg Mkc Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Uscg Mkc Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Uscg Mkc Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Uscg Mkc Study Guide
 - Personalized Recommendations

- Uscg Mkc Study Guide User Reviews and Ratings
- Uscg Mkc Study Guide and Bestseller Lists
- 5. Accessing Uscg Mkc Study Guide Free and Paid eBooks
 - Uscg Mkc Study Guide Public Domain eBooks
 - Uscg Mkc Study Guide eBook Subscription Services
 - Uscg Mkc Study Guide Budget-Friendly Options
- 6. Navigating Uscg Mkc Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Uscg Mkc Study Guide Compatibility with Devices
 - Uscg Mkc Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Uscg Mkc Study Guide
 - Highlighting and Note-Taking Uscg Mkc Study Guide
 - Interactive Elements Uscg Mkc Study Guide
- 8. Staying Engaged with Uscg Mkc Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Uscg Mkc Study Guide
- 9. Balancing eBooks and Physical Books Uscg Mkc Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Uscg Mkc Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Uscg Mkc Study Guide
 - Setting Reading Goals Uscg Mkc Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Uscg Mkc Study Guide
 - Fact-Checking eBook Content of Uscg Mkc Study Guide

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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