

1. Diffusion is a strong acid?
No acid that has all H⁺ ions dissociate (strong acids are considered)

2. Diffusion is a weak acid?
No acid

3. Strong acid?
Yes

4. Weak acid?
Yes

5. Diffusion is a strong acid?
No

6. Diffusion is a weak acid?
Yes

7. Diffusion is a strong acid?
No

8. Diffusion is a weak acid?
Yes

9. Diffusion is a strong acid?
No

10. Diffusion is a weak acid?
Yes

11. Diffusion is a strong acid?
No

12. Diffusion is a weak acid?
Yes

13. Diffusion is a strong acid?
No

14. Diffusion is a weak acid?
Yes

15. Diffusion is a strong acid?
No

16. Diffusion is a weak acid?
Yes

17. Diffusion is a strong acid?
No

18. Diffusion is a weak acid?
Yes

19. Diffusion is a strong acid?
No

20. Diffusion is a weak acid?
Yes

Acids and Bases: The Voyage of the Proton

Video found through Answering Lerner: [https://www.youtube.com/watch?v=...](#)

Statement and solution: Study and learn the voyage of the proton!

Dr. David Smith: Voyage of the Proton Answering Lerner

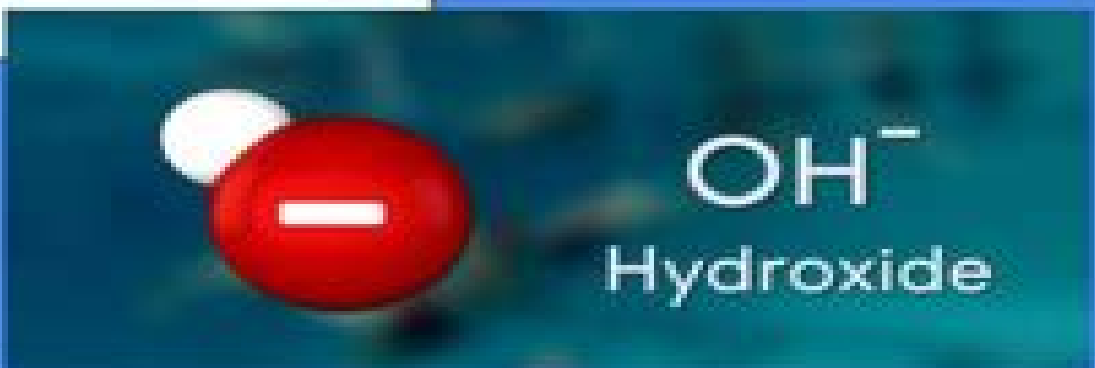
Questions/Answers: Before watching the video consider the following questions based on your knowledge of acids and bases.

1. What is the pH of a strong acid?
Answer: Low
2. What is the pH of a weak acid?
Answer: High
3. Can any acids of any acids and bases ever encounter a real life?
Answer: Yes

Questions: For the video, consider the following questions:

1. What is a hydronium ion? Is it a proton donor or proton acceptor?
H₃O⁺ is a proton donor (acid) and a proton acceptor (base).
2. What is a hydroxide ion? Is it a proton donor or proton acceptor?
OH⁻ is a proton acceptor (base) and a proton donor (acid).
3. What is a conjugate acid?
A molecule that can donate a proton.
4. What is a conjugate base?
A molecule that can accept a proton.
5. How do acids and bases affect pH?
Acids increase pH and bases decrease pH.
6. What is the pH of a neutral solution?
pH 7.
7. What is the pH of a strong acid?
Low pH.

KEY INCLUDED



Voyage Of The Proton Answers

**Dr Joseph Yeo, Teh Keng Seng, Loh
Cheng Yee, Ivy Chow**



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