Acids and bases end of unit worksheet

	Acids	Bases
Taste	Sour	bitter
oH range	0 - 7	7-14
Color with litmus paper	pink/red	blue
Are they electrolytes? (yes or no)	Ves	VES
What do they react with ?	carbonates bases	forts + oils
What ion do they produce?	H+ or H3O+	OH-
How do they feel?	(itch, burn)	Slippery
Color with phenolphthalein	colorless	Pink

- 2. I can list uses of acids in industry. List two uses of acids in industry. Cleaning concrete (HCI) Haking paint, plastics, fertilizers producing butteries
- 3. I can list examples of common acids and bases found in a home or store. List 2 common acids and 2 common bases you might find in your home or at a store. Acids: lemons, vinegar, sour candy, pop, aspirin, orange juice, ternonade

Bases: drain cleaner, Soap, baking soda, ammonia, 4. I can name acids and bases. Name the following:

HE hydrofluoric AcidHNO, Nitrous acid CHOH), calcium hydroxide

Write the formula for the following: (Check your charges!)

Magnesium hydroxide Mq(OH) > Hydrobromic acid HBC carbonic acid HaCO3

5. I can define and identify Arrhenius acids and bases. Label the following as an acid (A), base (B) or neutral (N).

6. I can label an acid as mono, di, or tri protic. Label the following as mono, di, or triprotic.

4,50, diprotic 4,00, triprotic 400, monoprotic HBr monoprotic

Weak Acids Bases Worksheet Answers

Mary Jones, Richard Harwood, Ian Lodge, David Sang

Weak Acids Bases Worksheet Answers:

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