

- 12 A sample of propanone, CH_3COCH_3 , was heated under reflux with potassium dichromate(VI) acidified with sulfuric acid, and then the mixture was distilled. Apart from the peaks due to the $\text{C}-\text{C}$ and $\text{C}-\text{H}$ bonds, what peak(s) would be present in the IR spectrum of the distillate?

- ☐ A A peak due to the $\text{C}=\text{O}$ only.
☐ B A peak due to the $\text{O}-\text{H}$ only.
☐ C Peaks due to $\text{C}=\text{O}$ and $\text{O}-\text{H}$.
☐ D Peaks due to $\text{C}-\text{O}$, $\text{C}=\text{O}$ and $\text{O}-\text{H}$.

(Total for Question 12 = 1 mark)

- 13 Which of the following describes the appearance of iodine under the stated conditions?

	Solid	Dissolved in aqueous potassium iodide	Dissolved in a liquid hydrocarbon
<input type="checkbox"/> A	purple	brown	purple
<input type="checkbox"/> B	brown	blue-black	yellow
<input type="checkbox"/> C	shiny grey	brown	purple
<input type="checkbox"/> D	shiny grey	brown	brown

(Total for Question 13 = 1 mark)

- 14 Why do calculations of global warming exclude the effect of water vapour in the atmosphere?

- ☐ A Water vapour is not a greenhouse gas.
☐ B Water vapour is a much less potent greenhouse gas than carbon dioxide.
☐ C The average concentration of water vapour in the atmosphere is fairly constant.
☐ D The concentration of water vapour in the atmosphere is much lower than that of carbon dioxide.

(Total for Question 14 = 1 mark)

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1. Understanding the eBook Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - The Rise of Digital Reading Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Advantages of eBooks Over Traditional Books
2. Identifying Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - User-Friendly Interface
4. Exploring eBook Recommendations from Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Personalized Recommendations
 - Unofficial Mark Scheme Chemistry Edexcel 2014 As User Reviews and Ratings
 - Unofficial Mark Scheme Chemistry Edexcel 2014 As and Bestseller Lists
5. Accessing Unofficial Mark Scheme Chemistry Edexcel 2014 As Free and Paid eBooks
 - Unofficial Mark Scheme Chemistry Edexcel 2014 As Public Domain eBooks
 - Unofficial Mark Scheme Chemistry Edexcel 2014 As eBook Subscription Services
 - Unofficial Mark Scheme Chemistry Edexcel 2014 As Budget-Friendly Options
6. Navigating Unofficial Mark Scheme Chemistry Edexcel 2014 As eBook Formats

- ePub, PDF, MOBI, and More
 - Unofficial Mark Scheme Chemistry Edexcel 2014 As Compatibility with Devices
 - Unofficial Mark Scheme Chemistry Edexcel 2014 As Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Highlighting and Note-Taking Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Interactive Elements Unofficial Mark Scheme Chemistry Edexcel 2014 As
 8. Staying Engaged with Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Unofficial Mark Scheme Chemistry Edexcel 2014 As
 9. Balancing eBooks and Physical Books Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Unofficial Mark Scheme Chemistry Edexcel 2014 As
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Setting Reading Goals Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Fact-Checking eBook Content of Unofficial Mark Scheme Chemistry Edexcel 2014 As
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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