



Name: _____ Date: _____
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Rule 1 A number is not needed to locate the double bond in unsaturated hydrocarbons with only one double bond because that bond is assumed to be between carbons 1 and 2.



cyclohexene



cyclopentene



1,3-cyclohexadiene



1,3-cycloheptadiene

Rule 2 In unsaturated hydrocarbons with only 1 double bond, the double bond is not shown on numbered 1 and 2 in the direction (priority or counterclockwise) that gives the first substituent.

Exception: In the case number 1, the number is used in the name to locate the double bond.



cyclohexene



1-methylcyclopentene



1-bromocyclohexene



1-ethyl-1-methylcyclohexene

Rule 3 In hydrocarbons with more than one double bond within the ring, assign one double bond the number 1 and use the other double bond the lowest possible number.



1,3-cyclohexadiene



1,3-cycloheptadiene

Alkenes as Substituents

An alkene group is a substituent hydrocarbon substituent in which a carbon-carbon double bond is present. The three most frequently encountered alkene groups are the vinyl, allyl, and isopropenyl groups, which may be named using suffixes: vinyloxy, allyloxy, methylenoxy, allyl-, and isopropenyl- or with common names: methoxy, ethoxy, prop-, and allyl-.

vinyl group allyl group isopropenyl group
ethoxy group methoxy group methoxy group
allyl group allyl group allyl group

Unsaturated Hydrocarbons Study Guide

Clyde Metz



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Table of Contents Unsaturated Hydrocarbons Study Guide

1. Understanding the eBook Unsaturated Hydrocarbons Study Guide
 - The Rise of Digital Reading Unsaturated Hydrocarbons Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Unsaturated Hydrocarbons Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Unsaturated Hydrocarbons Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Unsaturated Hydrocarbons Study Guide
 - Personalized Recommendations
 - Unsaturated Hydrocarbons Study Guide User Reviews and Ratings
 - Unsaturated Hydrocarbons Study Guide and Bestseller Lists
5. Accessing Unsaturated Hydrocarbons Study Guide Free and Paid eBooks
 - Unsaturated Hydrocarbons Study Guide Public Domain eBooks
 - Unsaturated Hydrocarbons Study Guide eBook Subscription Services
 - Unsaturated Hydrocarbons Study Guide Budget-Friendly Options
6. Navigating Unsaturated Hydrocarbons Study Guide eBook Formats

- ePub, PDF, MOBI, and More
- Unsaturated Hydrocarbons Study Guide Compatibility with Devices
- Unsaturated Hydrocarbons Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Unsaturated Hydrocarbons Study Guide
 - Highlighting and Note-Taking Unsaturated Hydrocarbons Study Guide
 - Interactive Elements Unsaturated Hydrocarbons Study Guide
- 8. Staying Engaged with Unsaturated Hydrocarbons Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Unsaturated Hydrocarbons Study Guide
- 9. Balancing eBooks and Physical Books Unsaturated Hydrocarbons Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Unsaturated Hydrocarbons Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Unsaturated Hydrocarbons Study Guide
 - Setting Reading Goals Unsaturated Hydrocarbons Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Unsaturated Hydrocarbons Study Guide
 - Fact-Checking eBook Content of Unsaturated Hydrocarbons Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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