# Understanding the Periodic Table

 Consolidate your understanding of the periodic table by using the word bank to complete the blanks below.

Word bank: periods protons noble gases alkaline earth metals groups synthetic atomic number halogens electrons bromine alkali metals metals mercury

Most elements in the Periodic Table occur naturally; however, some, such as technetium (Tc), rutherfordium (Rf), and fermium (Fm), are \_\_\_\_\_\_, meaning they were created in the laboratory.

Some groups in the Periodic Table have special names. Group 1 elements are known as the \_\_\_\_\_\_\_, Group 2 elements are called the \_\_\_\_\_\_\_, Group 17 elements are the \_\_\_\_\_\_\_, and Group 18 elements are the noble gases. Elements in the same group have the same number of \_\_\_\_\_\_\_ in their outer shell, which means they have similar chemical properties and tend to react in similar ways.

The \_\_\_\_\_\_ of an element increases from left to right across the Periodic Table. The atomic number represents the number of \_\_\_\_\_ present in the nucleus of an atom of that element. Each element has a unique atomic number.

Most elements in the Periodic Table are \_\_\_\_\_\_, some are nonmetals, and a few are metalloids. Metalloids are elements that display properties of both metals and nonmetals.

(Discrevelbreik)

# **Unscrambling The Periodic Table Answers**

Yan Bai

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using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es silicon Si and nitrogen N You don t need to be familiar
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$blanks\ in\ the\ puzzle\ For\ this\ puzzle\ the\ answer\ is\ P\ O\ S\ Es\ Si\ O\ N\ possession\ These\ VErBAl\ ReAcTiONS\ resemble\ chemical$
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example the chemical word POSSEsSiON is made using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es
silicon Si and nitrogen N You don t need to be familiar with the periodic table to solve these problems nor do you need to
know any chemistry You just need to be able to count and unscramble elements to make words This Hard volume consists of
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