

P390

[Total No. of Pages : 2

[4317] - 207
S.Y. B.Sc. (Semester - II)
CHEMISTRY
CH - 222 : Analytical Chemistry
(Paper - II) (2008 Pattern)

*Time : 2 Hours]**[Max. Marks : 40**Instructions to the candidates:-*

- 1) *All questions are compulsory.*
- 2) *Figures to the right indicate full marks.*
- 3) *Use of logarithmic tables and calculator is allowed.*
- 4) *Neat diagrams must be drawn wherever necessary.*

Q1) Answer the following :**[10]**

- a) How are samples of gases obtained?
- b) Which is a group reagent for V group?
- c) Define empirical formula.
- d) For which ions, EDTA can be used as a titrant.
- e) What is displacement titration?
- f) What are causes of randomness?
- g) How will you detect carboxylic group?
- h) Define solubility product.
- i) What is counter current extraction?
- j) Define equivalence point in volumetric analysis.

Q2) a) Answer any two of the following :**[6]**

- i) Derive the relation between the distribution coefficient and distribution ratio.
- ii) Explain the Borate removal scheme in inorganic qualitative analysis.
- iii) How 0.1 N iodine solution is prepared? Explain method for standardisation of iodine solution.

- b) Using kjeldahl method, the ammonia obtained from 0.5 gm of the sample was distilled into 50 ml of N/10 acid. The excess of the acid required 20 ml of N/10 base. Calculate the percentage of nitrogen. **[4]**

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