

1. Fill in the chart below

Name of element	Atomic number	# of protons	# of electrons	Category of element (metal, nonmetal, metalloid)	Valence electrons (go off column #) (no # larger than 8)	# of orbitals (period # - 10 #)	Bohr model of the element Draw Beryllium & Neon in the space below
Beryllium	4	4	4	Metal	2	2	
Silicon	14	14	14	metalloid	4	3	
Neon	10	10	10	Non-metal	8	2	
Aluminum	13	13	13	metal	3	3	
Carbon	6	6	6	non-metal	4	2	

2. Orbitals are also called energy levels

3. Most an atoms mass is found in the nucleus which is made up of protons & neutrons

4. ID the number of elements, molecules, & atoms in the compounds below

Compound	# of elements	Molecules	Element	atoms	Element	atoms	Element	atoms
$Fe(OH)_3$	3	7	Iron	7	Oxygen	14	Hydrogen	42
$Al_2(SO_4)_3$	3	4	Aluminum	12	Sulfur	12	Oxygen	48
$2MgO$	2	2	Magnesium	4	Chlorine	6		
$138 PO_4$	2	13	Phosphorus	39	Phosphorus	13	Oxygen	39
$MgSO_4$	3	1	Magnesium	1	Sulfur	1	Oxygen	4

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