

THEORY

EXPERIMENT

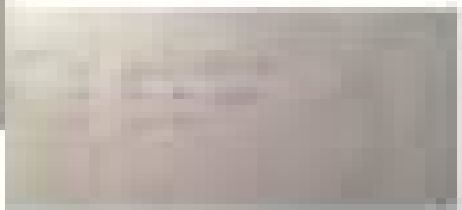
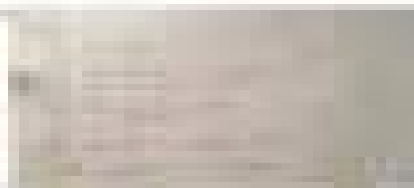
1. Introduction

The purpose of this experiment is to determine the rate of reaction between hydrogen peroxide and potassium iodide. The reaction is catalyzed by a small amount of potassium iodide. The rate of reaction is measured by the volume of oxygen gas produced over a period of time. The reaction is exothermic and the temperature of the reaction mixture increases as the reaction proceeds.

The reaction is as follows:

$$2H_2O_2 \rightarrow 2H_2O + O_2$$

The rate of reaction is determined by measuring the volume of oxygen gas produced over a period of time. The volume of oxygen gas is measured by the displacement of water in a graduated cylinder.



Tesccc Unit 07 Lesson 01 Key Mathematics

Camilla Rothe



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