MORE LIEUTE	-	
	Date:	Hour

# Information: Shapes of Molecules

Name	Methane, CH <sub>4</sub>	Ammonia, NH <sub>3</sub>	Water, H₂O
Lewis Structure	H	н—Т—н	H-,0,+
3-D Shape	Tetrahedral shape	Trigonal pyramidal shape  Bond angle = 106 5	Bent shape  Bond angle -104.5*
Total # of electron regions	4	4	4
# of Bonding electron regions	4	3	2
# of lone pair electron regions	0	1	2

Name	Carbonate, CO <sub>3</sub> <sup>2</sup>	Ozone, O <sub>3</sub>	Carbon dioxide, CO <sub>2</sub>
Lewis Structure	0,_0,_0	o#Ö~o	o=c=o
3-D Shape	Trigonal planar shape	Bent shape  Bent shape  Bend engle =118.6	Bond engle =180°
Total # of electron regions	3	3	2
# of bonding electron regions	3	2	2
# of lone pair electron regions	0	16	0

# **Test Molecular Shape Answer Key**

**Karin Nielsen-Saines** 

**Test Molecular Shape Answer Key:** 

The book delves into Test Molecular Shape Answer Key. Test Molecular Shape Answer Key is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Test Molecular Shape Answer Key, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Test Molecular Shape Answer Key
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