

Practice Final Exam

Pick the best answer for each of the following:

- _____ 1. Which of the following is NOT a true statement concerning what happens in a chemical reaction?
- The ways in which atoms are joined together are changed
 - New atoms are formed as products
 - The starting materials are called reactants
 - The bonds of the reactants are broken and new bonds of the products are formed
 - In a word equation representing a chemical reaction, the reactants are written on the left and the products on the right.

- _____ 2. In the balanced chemical equation for the following reaction, what is the coefficient in front of fluorine?



- a) 1 b) 2 c) 3 d) 5 e) 6

- _____ 3. In order to predict whether or not a redox (oxidation-reduction) reaction takes place, we

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- a) measure electron energies in eV
 b) orbital energy level diagram

- _____ 4. Predict the products for the following reaction:



- a) potassium zinc + nitrogen
 b) potassium nitride + zinc
 c) potassium nitrate + zinc
 d) potassium nitrate + zinc
 e) No Reaction

Lithium	<div style="display: flex; align-items: center;"> <div style="width: 10px; height: 100px; border-left: 1px solid black; border-right: 1px solid black; margin: 0 5px;"></div> <div style="writing-mode: vertical-rl; transform: rotate(180deg); font-size: 0.8em;">Increasing</div> </div>
Potassium	
Calcium	
Magnesium	
Zinc	

- _____ 5. Which of the following accurately represents the correct balanced equation for the following reaction:



- a) $5 \text{ S} + \text{N} \rightarrow \text{N}_5$
 b) $5 \text{ S} + \text{N} \rightarrow \text{N}_5\text{S}_5$
 c) $3 \text{ S}_8 + \text{N}_2 \rightarrow \text{N}_2\text{S}_6$
 d) $6 \text{ N}_2 + 5 \text{ S}_8 \rightarrow 2 \text{ N}_2\text{S}_6$
 e) $4 \text{ S}_8 + \text{N}_2 \rightarrow 2 \text{ N}_2\text{S}_8$

Texas Grade 1chemistry Final Practice Exam

Patrick Vollmar



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