

# THE FOLLOWING QUESTIONS ARE PRACTICE QUESTIONS FOR THE CSN CHEMISTRY PLACEMENT EXAM

The chemistry placement test is used to assess your preparation for General Chemistry I, CHEM 121 at CSN. The test is a standardized test that will examine your knowledge of chemistry and basic mathematical skills. You will be provided scratch paper, a Periodic Table of the Elements and a nonprogrammable scientific calculator. The following sample questions are intended only as a guide to show you the relative level of difficulty and example topics covered in the Chemistry Placement Exam. The actual exam will have a time limit of 45 minutes to answer 44 multiple choice questions. Click the following link to view a [Periodic Table](#).

**MULTIPLE CHOICE.** Choose the single choice that best answers the question.

- 1) How many total atoms are in the formula



A) 12                      B) 14                      C) 9                      D) 8

- 2) How many electrons are in  $\text{Br}^-$ ?

A) 4                      B) 7                      C) 36                      D) 34

- 3) What are the coefficients for the following reaction when it is properly balanced?



A) 2, 3, 6, 1                      B) 6, 1, 3, 2  
C) 2, 3, 1, 6                      D) 2, 1, 1, 3

- 4) Given that  $4 \text{ NH}_3 + 5 \text{ O}_2 \rightarrow 4 \text{ NO} + 6 \text{ H}_2\text{O}$ , when 4.50 mol of  $\text{H}_2\text{O}$  are formed, the amount of NO formed is

A) 3.00 mol                      B) 6.75 mol  
C) 1.50 mol                      D) 4.50 mol

- 5) What type of a reaction occurs when a sodium hydroxide solution is mixed with an acetic acid solution?

A) gas evolution  
B) oxidation-reduction  
C) acid-base neutralization  
D) precipitation

- 6) Which of the following atoms has the smallest radius?

A) Ne                      B) O  
C) Li                      D) B

- 7) What is the molecular geometry if you have a double bond, a single bond and 1 lone pair around the central atom?

A) trigonal pyramidal  
B) linear  
C) tetrahedral  
D) bent

- 8) What volume of 12.0 M HCl is required to make 75.0 mL of 3.50 M HCl?

A) 0.560 mL                      B) 257 mL  
C) 560. mL                      D) 21.9 mL

$$9) \frac{(2.80 \times 10^{-3})(1.28 \times 10^1)}{(3.20 \times 10^{-4})} = \underline{\hspace{2cm}}$$

A)  $1.27 \times 10^1$                       B)  $2.0 \times 10^2$   
C)  $1.2 \times 10^6$                       D)  $1.12 \times 10^2$

- 10) Of the molecules below, only \_\_\_\_\_ is polar.

A)  $\text{CH}_4$                       B)  $\text{SbF}_5$   
C)  $\text{CO}_2$                       D)  $\text{AsH}_3$

- 11) Which one of the following is a metalloid?

A) S                      B) Ge                      C) C                      D) Pb

- 12) A certain atom has a charge of -2 and has 16 protons and 16 neutrons in the nucleus. How many electrons does this atom possess?

A) 34                      B) 14                      C) 32                      D) 18

- 13) How many moles of pyridine ( $\text{C}_5\text{H}_5\text{N}$ ) are contained in 3.13 g of pyridine?

A) 0.00404                      B) 25.3  
C) 0.0396                      D) 0.319

- 14) Solve the following equation for x:

$$4x - b = 2x + c$$

A)  $c - b - 6$                       B)  $\frac{c + b}{2}$   
C)  $\frac{c + b}{6}$                       D)  $\frac{c - b}{2}$

# Texas Tech Chemistry Placement Exam Study Guide

**The Princeton Review, Marybeth  
Kravets, Imy Wax**



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**Course Success in the Undergraduate General Chemistry Lab** Thomas Elert, 2019-11-15 Stetig hohe Studienabbruchquoten in den MINT Fchern an deutschen Hochschulen welche auch aus geringem Kurserfolg in einf hrenden Laborpraktika resultieren k nnten und die wachsende Kritik an der Qualit t und Wirksamkeit ebendieser machen eine eingehende Betrachtung von Laborpraktika notwendig Diese Studie untersuchte die Lernziele des Laborpraktikums Allgemeine Chemie f r Lehramtsstudierende im ersten Semester sowie Faktoren f r den Kurserfolg um daraus Aussagen ber den Stellenwert von Laborpraktika in der universit ren Bildung insbesondere f r langfristigen Studienerfolg abzuleiten Dazu wurde ein theoretisches Modell zu Grunde gelegt welches das Vorwissen der Studierenden und die Lernzielpassung zwischen Studierenden und Lehrenden als zwei entscheidende Faktoren f r Kurserfolg ber cksichtigt Constantly high student dropout rates in STEM subjects at German universities which could be the result of low course success in introductory laboratory courses among other things and increasing criticism about their quality and effectiveness necessitate these laboratory courses to be examined thoroughly This study investigated the learning goals of the General Chemistry laboratory course for first year students in teacher training and factors for course success in order to make statements about the significance of laboratory courses for university education particularly for long term study success For this purpose a theoretical model that assumes the students prior knowledge and learning goal alignment between students and their lab instructors to be two defining factors for lab course success was used as a framework Resources in Education ,1999 The K&W Guide to Colleges for Students with Learning Differences, 16th Edition The Princeton Review, Marybeth Kravets, Imy Wax, 2023-09-05 FIND THE RIGHT SCHOOL FOR YOUR SPECIFIC NEEDS This indispensable resource will help students with ADHD Autism Spectrum Disorder or learning differences find and apply to their personal best fit college Hundreds of thousands of students with learning differences head to college every year This comprehensive guide makes it easy for those students and their families and guidance counselors to tackle the daunting process of finding the school that fits their needs best This invaluable book for students parents and professionals includes 350 school profiles with targeted information on admission requirements updated test policies and the support services and programs offered by the colleges Index of colleges by level of support Policies and procedures regarding course waivers and substitutions Resources to help students find the best match for their needs Advice from learning specialists on making an effective transition to college *Foundations of College Chemistry* Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with

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London from America in 1878 and proved himself an exceptional entrepreneur taking the pharmaceutical business by storm He was the brains and energy behind Burroughs Wellcome Co With his business partner Henry Wellcome he created an internationally successful firm the legacy of which can be found in the charity the Wellcome Trust yet few now remember him and the impact he made in his short lifetime A consummate salesman Burroughs was also an astute businessman with new ideas for marketing advertising and manufacturing his writings describe sales trips around the world and the people he met He was also a visionary employer who supported the eight hour working day profit sharing and numerous social and radical political movements including the single tax movement free travel Irish Home Rule and world peace In this first biography of Burroughs Julia Sheppard explores his American origins his religion and marriage and his philanthropic work as well as re

evaluating the dramatic deterioration of his relationship with his partner Wellcome      **Understanding Risk to Wildlife from Exposures to Per- and Polyfluorinated Alkyl Substances (PFAS)** Mark S. Johnson, Michael J. Quinn Jr., Marc A. Williams, Allison M. Narizzano, 2021-04-22 Understanding Risk to Wildlife from Exposures to Per and Polyfluorinated Alkyl Substances PFAS provides the most recent summary of toxicity data relevant to mammals birds reptiles and amphibians and provides values for use in risk assessment applications Predicting the bioaccumulation of PFAS in terrestrial wildlife including humans has proven to be extremely complex As a group PFAS act differently than traditional non ionic organic molecules where PFAS can break down and reform whereas some are demonstrated to be extremely persistent Where sufficient data are provided this book establishes toxicity reference values TRVs which are derived to assist in characterizing environmental sources of contamination and making risk based decisions Features Provides toxicity reference values TRVs for vertebrates mammals birds amphibians for PFAS where sufficient data are available and includes objective supporting background information Assigns a level of confidence to each TRV to provide the risk assessor with an understanding of the relative uncertainty associated with each value Presents toxicity data in the formats of scatter diagrams and tables for quick review and assessment Provides TRVs relevant for screening and decision making This book serves as a useful aid for risk assessors and managers in those industries that have sites contaminated with PFAS consultants tasked with evaluating risks at such sites and staff at regulatory agencies at various governmental levels who need to know how much contamination is considered safe for wildlife It will also appeal to researchers with an interest in filling the gaps in the current toxicological data for PFAS exposure

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