CHAPTER 5: ELECTRONS IN ATOMS

CHEMISTRY

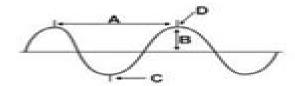
True/False

Indicate whether the statement is true or false.

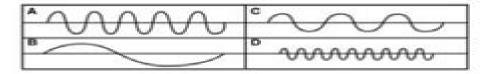
- There are a total of 5 d-orbitals.
 - The photoelectric effect describes the emission of electrons from a metal's surface when light at a specific frequency shines on it.
- Although the precise location of an electron cannot be determined, it is possible to determine the probability that an electron will occupy a certain region around the nucleus.
- 4. In the visible light spectrum, red light has a greater frequency (energy) than violet light.
- 5. [Kr]5s24d165p1 is the correct electron configuration for the element Bromine.
- 6. The maximum number of allowed valence electrons in any atom is 8.
- 7. Each element has its own unique emission spectrum.

Multiple Choice

Identify the choice that best completes the statement or answers the question.



- 8. Which label identifies the amplitude of the wave shown?
 - a. A c. C d. D
- Which label identifies the wavelength of the wave shown?
 a. A
 c. C
 - a. A e. C b. B d. D
- Which label identifies the trough of the wave shown?
 a. A
 c. C
 - b. B d. D



11. Which diagram shows a wave with the highest frequency?

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SJ Ball

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