

LESSON 17: CHEMICAL REACTION TYPES AND PREDICTING PRODUCTS

Student Learning Objectives

- Determine different chemical reaction types by looking at the reactants and products.
- Predict products of chemical reactions based on reaction types.

Key Vocabulary Terms

- Synthesis
- Decomposition
- Single Displacement
- Double Displacement
- Combustion
- Redox Reaction
- Acid/Base Reaction
- Precipitation Reaction



Key Equation

None... for this lesson.



17.1 Famous Five - Reaction Types

There are literally millions of different chemical reactions that occur every day throughout our world. Most of them occur without us realizing it. Scientists classify chemical equations into 5 specific groups. These groups help us determine what products may be produced from certain types of chemical reactions. We will look at synthesis, decomposition, single displacement, double displacement, and organic combustion reactions. Once we finish discussing these reactions, we will analyze some other types of chemical reactions which we can classify.

Synthesis reactions occur when two or more substances combine to form one new compound. They usually form by the general equation below.



A and B can be used to represent any element or compound. The above equation is just a general equation to show how the reactants become the specified products. Several examples of synthesis reactions are listed below.



Synthesis Reaction



Reactants

Product

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Austin Dulebohn



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Table of Contents Study Guide Types Of Reaction

1. Understanding the eBook Study Guide Types Of Reaction
 - The Rise of Digital Reading Study Guide Types Of Reaction
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide Types Of Reaction
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide Types Of Reaction
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide Types Of Reaction
 - Personalized Recommendations
 - Study Guide Types Of Reaction User Reviews and Ratings
 - Study Guide Types Of Reaction and Bestseller Lists
5. Accessing Study Guide Types Of Reaction Free and Paid eBooks
 - Study Guide Types Of Reaction Public Domain eBooks
 - Study Guide Types Of Reaction eBook Subscription Services
 - Study Guide Types Of Reaction Budget-Friendly Options

6. Navigating Study Guide Types Of Reaction eBook Formats
 - ePub, PDF, MOBI, and More
 - Study Guide Types Of Reaction Compatibility with Devices
 - Study Guide Types Of Reaction Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Study Guide Types Of Reaction
 - Highlighting and Note-Taking Study Guide Types Of Reaction
 - Interactive Elements Study Guide Types Of Reaction
8. Staying Engaged with Study Guide Types Of Reaction
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide Types Of Reaction
9. Balancing eBooks and Physical Books Study Guide Types Of Reaction
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide Types Of Reaction
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Study Guide Types Of Reaction
 - Setting Reading Goals Study Guide Types Of Reaction
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Guide Types Of Reaction
 - Fact-Checking eBook Content of Study Guide Types Of Reaction
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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