

- b) If the rate of consumption of magnesium is $5.0 \times 10^{-9} \text{ mol/s}$, find the rate of consumption of HCl in moles/s.

$$5.0 \times 10^{-9} \frac{\text{mol Mg}}{\text{s}} \times \frac{2 \text{ mol HCl}}{1 \text{ mol Mg}} = 1.0 \times 10^{-8} \text{ mol HCl/s}$$

Answer $1.0 \times 10^{-8} \text{ mol HCl/s}$

- c) If the rate of consumption of magnesium is $5.0 \times 10^{-9} \text{ mol/s}$, find the rate of production of H_2 in g/s.

$$5.0 \times 10^{-9} \frac{\text{mol Mg}}{\text{s}} \times \frac{1 \text{ mol H}_2}{1 \text{ mol Mg}} \times \frac{2.0 \text{ g H}_2}{1 \text{ mol H}_2} =$$

Answer $1.0 \times 10^{-8} \text{ g H}_2/\text{s}$

- d) If the rate of consumption of magnesium is $5.0 \times 10^{-9} \text{ mol/s}$, find the rate of production of H_2 in L/s (@STP).

$$5.0 \times 10^{-9} \frac{\text{mol Mg}}{\text{s}} \times \frac{1 \text{ mol H}_2}{1 \text{ mol Mg}} \times \frac{22.4 \text{ L H}_2}{1 \text{ mol H}_2} =$$

Answer $1.1 \times 10^{-7} \text{ L H}_2/\text{s}$

- e) If the rate of consumption of magnesium is $5.0 \times 10^{-9} \text{ mol/s}$, find the mass of Mg consumed in 5.0 minutes.

$$5.0 \times 10^{-9} \frac{\text{mol Mg}}{\text{s}} \times \frac{24.3 \text{ g Mg}}{1 \text{ mol Mg}} \times \frac{60}{1} \times 5.0 \text{ min} = 3.6 \times 10^{-5} \text{ g Mg}$$

Answer $3.6 \times 10^{-5} \text{ g Mg}$

3. When butane (C_4H_{10}) is burned in air (oxygen), the products carbon dioxide and water are formed.

- a) Write a balanced formula equation for this reaction.



- b) If butane is consumed at an average rate of 0.116 grams/s, determine the rate of production of CO_2 in g/s.

$$0.116 \frac{\text{g C}_4\text{H}_{10}}{\text{s}} \times \frac{1 \text{ mol C}_4\text{H}_{10}}{58.1 \text{ g C}_4\text{H}_{10}} \times \frac{8 \text{ mol CO}_2}{2 \text{ mol C}_4\text{H}_{10}} \times \frac{44.0 \text{ g CO}_2}{1 \text{ mol CO}_2} = 0.352 \text{ g CO}_2/\text{s}$$

Skill Practice 34 Percent Yield Answers

David A. Dieterle



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In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Skill Practice 34 Percent Yield Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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