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Teacher:	
Date:	
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References

Chemistry in the Community pp 45-65 Water properties concept map Solubility concept map Sclubility, graphing and analysis Solubility Practice: A Curvy Subject. Study Solvent Lab

Vocabulary

Understand and apply the definitions or symbols of the following terms.

Colligative properties Seduction Distillation Subsect Filtration: Selution Homogeneous. Supersaturated. Militarium est. Saturated Nomoolian Unsuturated: (Prober) See good Se-Pure substances. Solubility

Targeted Skills

- Describe the structure of water and relate its unusual physical properties to its structure, high bolling point, surface tension, high specific heat.
- Draw the structure of water and relate its structure to the ability of water to act as a universal. sundynamit.
- Compare/contrast the solubilities of various ionic and covalent solutes.
- 4. Apply: Like dissolves like:

Potar solvents dissolve Potar and lonic solutes.

Nonpolar solvents dissolve Nenpolar solutes

- Describe factors that affect the rate of dissolving, such as particle size, stiming, temperature, nature of the solid solutes.
 - Interpret information from a solubility curve (graph).

Make sure you remember from past tests?

Atomic structure Significant figures Accuracy & precision

Equations:(extension)

Solution concentration = amount of solute

amount solute + solvent

Test Breakdown

30 Multiple-Choice Problems (max.) - 60%

- 1 Free Response Problem chosen from the Free Response Test Bank 10%
- 1 Open-Ended Problem chosen from the Open-Ended Test Bank 10%
- 1 Performance Assessment 20%

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Coates, Mary-Margaret Coates, Jan Riordan

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