

Chemistry Spring Final Study Guide

Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
Common	Common

Unit 10: Acid-Base

Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below.

Label the pH scale.

Name the three categories of acids.

1. Strength of the particles
2. Strength of the particles
3. A number of
4. The gas in it
5. The number

Unit 10: Acid-Base

Write the definition:

- Oxidation
- Reduction
- Reduction
- Oxidation
- Reduction
- Oxidation

Use the equation that:

Name the correct for:

1. When a base reacts with an acid
2. When a base reacts with an acid

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Name: _____

Unit 1: Measurement Analysis

Name the Unit Symbols from these formulas to make accurate.

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Determine the number of Significant Figures in each of the following numbers:

Mass = 1.23456789	Mass = 1.23456789
Volume = 1.23456789	Volume = 1.23456789
Temperature = 1.23456789	Temperature = 1.23456789

Convert the following into Scientific Notation:

Mass = 1.23456789	Mass = 1.23456789
Volume = 1.23456789	Volume = 1.23456789
Temperature = 1.23456789	Temperature = 1.23456789

Unit 2: Chemical Reactions

Name the Unit Symbols from these formulas to make accurate.

Name the Unit Symbols from these formulas to make accurate.

Determine the number of Significant Figures in each of the following numbers:

Mass = 1.23456789	Mass = 1.23456789
Volume = 1.23456789	Volume = 1.23456789
Temperature = 1.23456789	Temperature = 1.23456789

Convert the following into Scientific Notation:

Mass = 1.23456789	Mass = 1.23456789
Volume = 1.23456789	Volume = 1.23456789
Temperature = 1.23456789	Temperature = 1.23456789

Unit 3: Reaction

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

Compound	Molar Mass	Percent Composition
H ₂ O		
NaCl		
NaOH		
Na ₂ O		
Na ₂ CO ₃		

Write the chemical formula for each compound:

Compound	Chemical Formula
Hydrogen	
Oxygen	
Carbon	
Nitrogen	
Sulfur	

Write the chemical formula for the compound:

Diagram of a chemical reaction:

```

graph TD
    A[Reactants] --> B[Products]
    B --> C[Energy]
    C --> D[Energy]
    
```

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**New Mexico College of Agriculture and
Mechanic Arts, New Mexico State
University**

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Catalog and Announcement of North Dakota Agricultural College for ..., 1924 Resource Series Midwest
Program on Airborne Television Instruction, 1963 *Catalog* Midland Lutheran College, 1895 **Bulletin** Northern
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Junior College, Santa Barbara State Normal School of Manual Arts and Home Economics, University of California, Santa
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